PALAEOPROTEROZOIC PROGRADE METASOMATIC-METAMORPHIC OVERPRINT ZONES IN ARCHAEAN TONALITIC GNEISSES, EASTERN FINLAND

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Several occurrences of coarse-grained kyanite rocks are exposed in the Archaean area of eastern Finland in zones trending predominantly northwest-southeast that crosscut all the Archaean structures and, locally, the Palaeoproterozoic metadiabase dykes, too. Their metamorphic history illustrates vividly Palaeoproterozoic reactivation of the Archaean craton.

The early-stage kyanite rocks were formed within the framework of ductile shearing or by penetrative metasomatism in zones of mobile brecciation. Static-state coarse-grained mineral growth during the ongoing fluid activity covered the early foliated fabrics, and metasomatic zoning developed. The early-stage metasomatism was characterized by Si, Ca and alkali leaching. The late-stage structures are dilatational semi-brittle faults and fractures with unstrained, coarsegrained fabrics often formed by metasomatic reactions displaying Mg enrichment along grain boundaries. Metamorphism proceeded from the low-T earlystage Chl-Ms-Qtz, Ky/And-St, eventually leading to the high-T late-stage Crd-Sil assemblages. The thermal peak, at 600-620°C/4-5 kbar, of the process is dated to 1852±2 Ma (U-Pb) on xenotime. Al-silicate growth successions in different locations record small variations in the Palaeoproterozoic clockwise P-T paths. Pressure decreased by c. 1 kbar between the early and late stage, i.e. some exhumation had occurred. Fluid composition also changed during the progression, from saline H₂O to CO₂ rich. Weak retrograde features of high-T phases indicate a rapid cooling stage and termination of fluid activity.

The early-stage Ky-St assemblages resemble those described from nearby Palaeoproterozoic metasediments in the Kainuu and North Karelia Schist Belts, where the metamorphic peak was achieved late with respect to Palaeoproterozoic structures. The static Ky-St metamorphism in kyanite rocks was generated by fluid-induced leaching processes at elevated T during the post-orogenic stage after collision of the Palaeoproterozoic island arc complex with the Archaean craton in Palaeoproterozoic and/or reactivated Neoarchaean zones of weakness. The distribution of late-stage Crd-Sil metasomatism-metamorphism indicates that the corresponding thermal event was widespread in the Nurmes–Sotkamo area. Isotopic studies on Archaean granitoids and greenstone belts also indicate such late heating. According to pressure determinations, the Archaean±Palaeoproterozoic crust has been uplifted and exhumed about 15–20 km since 1850 Ma. Contemporaneous magmatic activation occurred in the North Karelia Schist Belt, too. The seismic deep structure of the crust, particularly the thick high-velocity layer in the lower crust, postulates some kind of disturbance in the lower crust

and lithospheric mantle. The dilatational late-stage heat flow, CO_2 -rich fluidization and Mg metasomatism, and exhumation are connected with this disturbance. It is suggested that they were related to magmatic underplating into the lower crust after the Svecofennian-Archaean collision.

Key words: Archean, Paleoproterozoic, tectonics, reactivation, overprint, metamorphism, metasomatism, P-T path, mineral reaction, fluids, fluid inclusions, kyanite, cordierite, xenotime, U/Pb age, eastern Finland

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INTRODUCTION

The Meso- to Neoarchaean craton of eastern Finland, the western part of the large Archaean domain of the Fennoscandian Shield, consists of tonalitic-trondhjemitic-granodioritic migmatites and granitoids (TTG) and Neoarchaean greenstone and paragneiss belts (Simonen 1980, Kontinen 1991, Luukkonen 1992). It is bordered by the Palaeoproterozoic Svecofennian Domain along a suture zone in the southwest (Koistinen 1981). The southwestern margin of the craton is partly covered by overthrust allochthonous and autochthonous Palaeoproterozoic sequences (Koistinen 1981, Huhma 1986) (Fig. 1). A detailed structural analysis of the craton in the Kuhmo area and its evolutionary history were presented by Luukkonen (1992, see also references therein). The few metamorphic studies conducted show that amphibolite- and, locally, granulite-facies conditions were achieved in the TTG areas and greenschistto amphibolite-facies conditions in the greenstone belts. Widespread low-grade retrogression characterizes both associations (Taipale 1982, Paavola 1984, Tuisku & Sivonen 1984, Hölttä & Paavola 1989, Blais & Auvray 1990, Gruau et al. 1992).

The Palaeoproterozoic overprint on the Archaean craton has aroused increasing interest in recent years. The Palaeoproterozoic extensional rifting stages are recorded by several mafic magma pulses between 2.44 and 1.97 Ga ago (Aro & Laitakari 1987, Alapieti & Lahtinen 1989, Vuollo 1994). About 1.9-1.875 Ga ago the Svecofennian island arc complex collided with the passive Archaean craton margin (Vaasjoki & Sakko 1988). The effects of this collision, seen in the nappes and overthrusts northeast of the suture zone, were discussed first by Wegmann (1928) and Väyrynen (1939). Park and Bowes (1983) and Park (1988) studied the basement-cover relations during Palaeoproterozoic polyphase deformation in the contact area of the North Karelia Schist Belt and the Archaean craton, noting the sliced structure of the Archaean basement owing to Palaeoproterozoic tectonism but preservation of Archaean structures further away from the thrust zones. Recent studies have confirmed that the Svecofennian orogeny gave rise to the formation of a block structure in the Archaean area, too (e.g. Kohonen et al. 1991, Kohonen 1995). Jegouzo and Blais (1993) observed that a continuation of some Palaeoproterozoic structures of the Kainuu Schist Belt overprints the Archaean basement in the Sotkamo area, and that Palaeoproterozoic diabase dykes are overprinted by metamorphic schistosity near the Kuhmo Greenstone Belt. The steeply foliated Palaeoproterozoic sequences within the Archaean in central Russian Karelia, east of the Finnish border, demonstrate the considerable extent of the Palaeoproterozoic reconstructive structures in the eastern Archaean area (see Ward 1993). To the west, however, only a weak Palaeoproterozoic deformation in the Lentiira and Ilomantsi areas has been described (Luukkonen 1985, 1988, 1989a, Ward



Fig. 1. Kyanite rock occurrences (black dots = exposed; boulder = local boulders) in eastern Finland: 1 = Tuomaanvaara, 2 = Turkkivaara, 3 = Kallioniemi, 4 = Varisniemi, 5 = Korpilampi, 6 = Kauhealampi, 7 = Havukkalammit and east Havukkalammit, <math>8 = Matovaara, 9 = Levävaara, 10 = Hiltuspuro, 11 = Teljo, 12 = Egyptinkorpi, 13 = Lehtovaara (not studied), 14 = Tetrilampi, 15 = Kynsiniemi (not studied), 16 = Nilsiä (not studied). Hatched lines = tectonic boundaries: (H) = Nunnanlahti–Holinmäki Shear Zone and (S) = approximate location of the suture zone between Svecofennian island arc complex and Archaean complexes; thin dashed line = border between the Nurmes–Sotkamo and Lieksa areas, defined on basis of geophysical data.

1993). On the basis of geophysical data, Korja et al. (1993) suggested that the lower crust and Moho of the westernmost Archaean was also reformed during Palaeoproterozoic time.

Isotopic resetting is recorded in the K-Ar isotopic compositions of TTG biotite and hornblende, which show late heating to temperatures exceeding the biotite and hornblende K-Ar closure temperatures, that is, about 300°C and 500°C, respectively, over a large area of the Archaean. Biotites were reset, on average, 1795±21 Ma ago and hornblendes 1851±41 Ma ago (Kontinen et al. 1992). O'Brien et al. (1993) determined K-Ar ages of 1811-1707 Ma on biotites in the Ilomantsi area, and attributed them to Palaeoproterozoic metamorphism; Karhu et al. (1993) made a similar suggestion on the basis of the large variation in the oxygen isotope composition of carbonates in that area. According to Gruau et al. (1992), the Nd and O isotopic and REE compositions of the southern Kuhmo Greenstone Belt komatiites were modified by regional metamorphism and CO₂ fluidization about 1.8 Ga ago. Medium-pressure kyanite-sillimanite metamorphism is widespread in the eastern Archaean terrain (Korsman 1982, Russian Academy of Science et al. 1993). Belyaev et al. (1993) and Petrov (1993) among others, attributed it to Palaeoproterozoic reactivation.

The main constraint on metamorphic studies in the Archaean area is the scarcity of appropriate mineral parageneses for conventional thermobarometric studies. In several locations, however, kyanite-bearing rocks have served as valuable indicators of metamorphic crystallization conditions. Here, we focus on the zones of spectacular, coarse-grained kyanite rocks that occur in the TTG environment. These rocks have a wide diversity of composition and fabric. Previous investigations concentrated on kyanite and U-Th prospecting, without much emphasis on the origin of these kyanite-bearing rocks (Rask 1979, Pekkala 1982, Äikäs 1989, Horneman & Hyvärinen 1987, 1989).

Our study is based on observations made at several kyanite rock sites – the term *kyanite rock* refers to the rock association in a general sense – in the Nurmes–Sotkamo area (Fig. 1). We describe kyanite rocks in detail and examine their tectonic-metamorphic evolution and genesis and their relation to the Palaeoproterozoic reactivation of the Archaean crust. Petrological study is supported by fluid inclusion data. The tectonic-metamorphic succession was dated with structural and isotopic methods.

This research is part of the "Global Geoscience Transects" project of the Geological Survey of Finland. The fluid inclusion determinations, comprising microthermometry and studies of composition and morphology, were made by Matti Poutiainen. Matti Pajunen is responsible for all the rest including the interpretations presented here.

GEOLOGICAL OUTLINE

The Archaean craton of eastern Finland can be divided into the Iisalmi granitoid-migmatite area to the west of the Palaeoproterozoic Kainuu Schist Belt with its structural extension in the Nunnanlahti-Holinmäki Shear Zone (Kohonen et al. 1991), and the eastern Archaean area. Eastwards the thickness of the crust decreases dramatically, from 56 to 46 km, in the vicinity of the Kuhmo Greenstone Belt (Yliniemi et al. 1993). This decline in crustal thickness is due to the thinning of the highvelocity layer of the lower crust east of the Kuhmo Greenstone Belt (Korja et al. 1993, Korja et al. 1994). The granitoid-migmatite area between the Kainuu Schist Belt and the Kuhmo Greenstone Belt is called the Nurmes-Sotkamo area and the area of thin crust to the east of the Kuhmo Greenstone Belt, the Lentiira area. The Lieksa area (Fig. 1) is defined on the basis of its characteristic magnetic and gravimetric pattern only.

There are several records of rocks over 3.0 Ga old (e.g. Paavola 1986, Kozhevnikov et al. 1987, Sorjonen-Ward & Claoué-Long 1993) from the Archaean area of eastern Finland and Russian Karelia. According to structural observations of Luukkonen (1988, 1992), the oldest remnants of the Archaean crust in eastern Finland are the highly deformed amphibolites in trondhjemitic-tonalitic migmatites. The early mafic to intermediate crust was deformed by D_1 and D_2 , and was subject to dynamothermal medium- to high-grade metamorphism about 2843±18 Ma ago (Luukkonen 1985).

Intracontinental rifting about 2.79–2.7 Ga ago produced ultramafic to felsic metavolcanics and volcanoclastic metasediments of the greenstone belts (Luukkonen 1992). Widespread paragneisses in the Nurmes–Sotkamo area have a calc-alkaline volcanic-plutonic provenance (Kontinen 1991). The rocks of the early TTG crust and greenstones were tightly to isoclinally folded, overthrust, fragmented by D_3 and intruded by felsic D_3 magmas. Luukkonen (1992) connects the thermal event with magmatic underplating. Unlike the amphibolite or greenschist facies metamorphism present in most of the terrain, granulite facies Archaean metamorphism is restricted to sharply defined areas, e.g. in the Iisalmi (Paavola 1984, Hölttä & Paavola 1989) and Lieksa areas (Kontinen & Paavola 1996), and in Russian Karelia near the Finnish-Russian border (Kozhevnikov et al. 1987, Belyaev et al. 1991).

Late Neoarchaean structures record cratonization of the crust. D_4 structures, northwest-southeast and conjugate northeast-southwest-trending faulting and folding, indicate diminishing ductility under retrograde conditions. Aplogranitic neosome in F_4 axial planes has been dated to 2657±32 Ma. The latest Neoarchaean D_5 and D_6 structures appear as open folds. Felsic intrusives in D_6 fractures have been dated to c. 2642–2575 Ma (Luukkonen 1985).

The 2.44-Ga mafic layered intrusions and dykes (Alapieti & Lahtinen 1989, Vuollo 1994) and 2.2-1.97-Ga diabase magmatism (Aro & Laitakari 1987, Vuollo 1994) imply tensional early Palaeoproterozoic conditions resulting in break-up of the Archaean "supercontinent" (see Lahtinen 1994). Most of the diabase dykes trend east-west and northwest-southeast (Luukkonen 1985, Aro & Laitakari 1987) and, according to Luukkonen (1985), were often intruded along reactivated Archaean zones of weakness. The Archaean structures are crosscut by anorogenic 2435±12 Ma (Luukkonen 1989b) to 2352±25 Ma (Horneman 1990) old granites. These earliest Palaeoproterozoic processes were followed by collision of the Svecofennian domain with the Archaean craton from the south and southwest (Koistinen 1981). The collisional and post-collisional events were outlined in section "Introduction"

KYANITE ROCKS

The majority of the known kyanite rock occur-

rences in the eastern Archaean area are in the Nurmes–Sotkamo area (Fig. 1). Similar rocks are exposed in a zone trending north–south through Nilsiä in the Iisalmi area. Despite detailed geological mapping north of Kuhmo, no corresponding occurrences have been found there; nor have such rocks been encountered in the Lieksa area. At most of the sites studied, the contact between the kyanite rock and its host is well exposed. The structures and mineral growth history are particularly well exposed at Hiltuspuro, Nurmes, which is why this site was chosen for detailed study.

Hiltuspuro

Country rocks

At Hiltuspuro the kyanite rock is exposed in a $4 \times$ 0.5 km² area associated with a lineament trending northwest-southeast on the low-altitude aeromagnetic map and a long valley on the topographic map. The country rocks are tonalites and tonalitic migmatites with amphibolitic and mica gneissic remnants (Figs. 2a and b). Amphibolite has a regional metamorphic assemblage of Pl-Hbl± Qtz±Grt (mineral abbreviations in Appendix 1); only tiny garnets are found in the Hiltuspuro area. Structurally, the tonalites are comparable to the D_3 granitoids of Luukkonen (1992); they are homogeneous and medium grained with a metamorphic microstructure and the mineral assemblage, Pl-Qtz-Bt±Hbl±Ksf (Appendix 2). Medium-grained and pegmatitic potassic granite dykes crosscut the other rock types. Small retrograde and sheared ultramafic bodies occur in the southeastern part of the area. In northwest Hiltuspuro a coarse-grained garnet-quartz rock is associated with a kyanite rock. The rotated internal structure of some garnet grains suggests growth under simple shear conditions. A few small garnets are partly altered to chlorite and rimmed by a younger-generation garnet.

Diabase dykes up to $100x500 \text{ m}^2$ in size (see Fig. 2b), similar to the dated Palaeoproterozoic dykes, are weakly to moderately foliated. The mineral lineation is well developed in narrow dykes although, a blasto-ophitic microstructure is



Fig. 2. Lithology and tectonic features in the Hiltuspuro area: a) northwest Hiltuspuro, b) southeast Hiltuspuro.



often still visible. The primary mineral assemblage has completely recrystallized as an amphibolite facies assemblage, Hbl-Pl-Ilm/Mag. Epidotization and biotitization are distinct in some strongly foliated dykes.

Deformation

The earliest Archaean structure identified at Hiltuspuro (Figs. 2a and b) is a penetrative S_{1-2} mineral foliation in amphibolitic remnants that has been folded by tight to isoclinal F₃ plunging moderately northwards. The composite foliation is often the only detectable structure in amphibolites. In tonalites the penetrative mineral foliation is northerly to northeasterly S₃. This foliation has been zonally reoriented by open to tight F4 folds plunging moderately northwestwards or, rarely, to southeast. Weak S₄ dips steeply or moderately to southwest. The sheared limbs of F₄ folds are characterized by pegmatitic granite and biotitization of amphibolite. These structures broadly correspond to those described by Luukkonen (1988) from further north.

The observed contacts of the Palaeoproterozoic metadiabase dykes trend northwest-southeast and are sheared. Lineation L_{P1} (P1 = Palaeoproterozoic structure in metadiabase) plunges moderately northwestwards, and mineral foliation S_{P1} is subvertical, dipping steeply southwestwards. Thus, there is no marked difference between the trends of the regional Neoarchaean D_4 (see Luukkonen 1988) and D_{P1} . In fact, exact identification of the Neoarchaean D_4 and the Palaeoproterozoic D_{P1} in the tonalitic country rocks is difficult except near the dyke contacts, where S_{P1} may occasionally be traced to the country rock.

The penetrative Neoarchaean structures of tonalites are crosscut by breccia, fault and fracture zones. The breccia zones consist of a coarsegrained, strongly recrystallized rock with indistinct mica foliation shown by later static mineral overgrowth. The breccia zones have weakly sheared, gradual or even "intrusive" replacement contacts (see a corresponding structure in Fig. 12 from Levävaara) with tonalite. Tonalitic fragments are corroded and assimilated by coarse-grained brec-

cia matrix. Weak northwest-southeast crenulation deforms the zones. The surroundings of the breccia zones are characterized by semi-brittle, curved fractures and faults, up to 10 cm wide, that penetrate the Neoarchaean structures from northwest to southeast and locally from northeast to southwest and north to south. This structure dips steeply or moderately to southwest. A predominantly dextral strike slip component is evident in the northwest-southeast-trending faults, but the existence of an opposite displacement, too, makes interpretation of the sense of displacement difficult, as good vertical sections for displacement observations are rare. Indeed, the shear sense can be seen only along the contact zones; their interior, however, is characterized by unoriented, static mineral growth, as in the coarse-grained zones described above.

The orientation of the semi-brittle structure is close to that of the regional Neoarchaean D_4 . However, D_4 is compressional/transcompressional folding and ductile in character, whereas the semi-brittle zones typically show dilatational features such as filled fractures. The semi-brittle zones crosscut the syn- D_4 pegmatitic granite dykes and the Palaeoproterozoic metadiabase dykes, in which the metamorphic assemblage, Hbl-Pl-Ilm/Mag, has altered to Hbl-Ep-Ab-Qtz. Accordingly, the semibrittle deformation is younger than the Palaeoproterozoic D_{P1} , and is named here D_{P2} (= deformation later than D_{P1}).

Kyanite rock

In the late breccia and semi-brittle D_{P2} zones, the Archaean tonalites have altered to heterogeneous kyanite rocks. The changes in mineral composition involve differential mobility of the primary rock-forming components and a metasomatic alteration process (cf. Korzhinskii 1970). Close to the kyanite rock the primary mineral assemblage of tonalite has altered to Pl-Qtz-Chl-Bt-Rt through decomposition of biotite to chlorite and rutile. Hornblende, K-feldspar and sphene are absent. Thus, the tonalite is leucotonalitic in appearance and is characterized by narrow fractures filled with chlorite and quartz and by irregular chlorite-rich patches. Plagioclase is often coarser within chlo-



Fig. 3. Coarse-grained kyanite-chlorite-cordierite rock at Hiltuspuro. Gradual metasomatic zoning is expressed by kyanite- (centre), plagioclase- and chlorite-biotite-rich (left) portions. The photograph is from the outcrop shown in Fig. 2b and sketched in Fig. 5. Photo by M. Pajunen.



Fig. 4. Kyanite-cordierite association in semi-brittle dilatational fractures, D_{P2} , in tonalite at Hiltuspuro. Biotite aggregates form spots, and tonalite is enriched in felsic components in vicinity of fractures. Photo by M. Pajunen.

ritized portions of the rock than in unaltered tonalite.

The altered rocks are either (1) coarse-grained kyanite-chlorite-cordierite rocks in breccia zones (Fig. 3), (2) kyanite-cordierite rocks in semi-brittle fracture/fault zones (Fig. 4) or (3) irregular cordierite-rich patches replacing the tonalite without distinct deformation on outcrop scale.

The ultimate metasomatism of tonalite is expressed by a coarse-grained kyanite-chlorite-cordierite rock that crops out in southeast Hiltuspuro (Fig. 2b). A boulder, a few cubic metres in size, in northwest Hiltuspuro (Fig. 2a) implies more extensive distribution of this

penetrative process. The chlorite- and biotiterich groundmass of the rock is overgrown by unoriented kyanite laths (average 5.5–6.5% kyanite in some outcrops, Rask 1979), up to 15–20 cm long, scattered grains of staurolite, 1 cm across, fresh, blue cordierite and scarce fibrous sillimanite. This quartz-free rock shows coarse, irregular metasomatic zoning (Fig. 5).

A coarse-grained, entirely recrystallized chlorite-staurolite-plagioclase rock with staurolite porphyroblasts, 1–3 cm across, in southeast Hiltuspuro replaces the tonalite with irregular contacts. It is similar to the kyanite-chlorite-cordierite rock in appearance, and con-



Fig. 5. Field sketch of coarse-grained kyanite-chlorite-cordierite rock with metasomatic zoning at Hiltuspuro. Location of outcrop shown in Fig. 2b.



Fig. 6. Annealed plagioclase, Pl_2 , and quartz showing crystallization in unstrained state in semi-brittle fracture (to the right) at Hiltuspuro. Bt and Pl_1 represent relics of "primary" minerals in tonalite. Field of view c. 4 mm wide. Crossed nicols. Photo by M. Pajunen.

tains irregular staurolite-, chlorite- and plagioclase-rich portions.

- (2) The Archaean foliation in tonalite was reoriented by D_{P2} close to the semi-brittle kyanitecordierite rock zones. Plagioclase recrystallized and biotite accumulated into spots. In the margin of the zone, the plagioclase and quartz are smaller and poorly annealed (Fig. 6). In the central part of the zones, metasomatic neomineralization took place under static conditions as evidenced by the undeformed fabric. Random grains of kyanite 1–5 cm in diameter, and of staurolite, up to 0.5 cm in diameter, are surrounded by cordierite and finegrained sillimanite. Chlorite is a minor phase in these zones.
- (3) Cordierite-rich patches, 5–30 cm in diameter, and irregular zones, up to 2–3 m wide, replace the tonalite without a clear link to deformation on outcrop scale. In places, however, the zones follow the direction of the earlier foliation – Neoarchaean S₄ or Palaeoproterozoic

 D_{P1} . Under the microscope the overall replacement of the previous phases by cordierite± sillimanite and quartz is pronounced, and the early phases are preserved only as small relics in large cordierite porphyroblasts (Fig. 7). In northwest Hiltuspuro, fibrous, radial orthoamphibole aggregates, a few millimetres in diameter, occur in cordierite. Coarse-grained biotite and chlorite form large aggregations, especially within cordierite-rich rocks.

The succession of mineral growth is remarkably systematic in the altered rocks. The metamorphic-metasomatic textures are described in greater detail in section "Metasomatic-metamorphic reaction history" (pp. 93–104). Various, dominantly quartz-free biotite-, chlorite- and plagioclase-bearing early parageneses are well preserved in the coarse-grained kyanite-chlorite-cordierite rock. The mica-rich assemblage is overgrown by static kyanite and staurolite that were crystallized contemporaneously.



Fig. 7. Detail of cordierite-rich patch including remnants of "primary" plagioclase and recrystallized quartz at Hiltuspuro. Field of view c. 4 mm wide. Crossed nicols. Photo by M. Pajunen.

The early, "low-T" kyanite-staurolite paragenesis is overprinted by later, "high-T" cordierite and contemporaneous sillimanite, with the result that kyanite and staurolite occur as corroded relics in cordierite (Fig. 8). Biotite has decomposed into cordierite in contact with kyanite. Cordierite replaces the other phases, even plagioclase, in kyanite rocks. Fibrous sillimanite was formed by, for instance, polymorphic transformation of kyanite. Orthoamphibole crystallized after chlorite and biotite. In general, the semi-brittle zones and cordierite-rich patches show mineral assemblages representing more advanced metasomatism under late "high-T" conditions.

Retrogressive metamorphic reactions, e.g. pinitization of cordierite, muscovitization of aluminium silicates, muscovitization and chloritization of staurolite, and carbonatization of orthoamphibole, were so weak that prograde parageneses are always dominant. These reactions are often restricted to the surroundings of late fractures.

Other targets

The kyanite rocks close to the Kainuu Schist Belt (Fig. 1) are connected with the shear zones trending northwest-southeast and dipping southwest. Early features are well preserved at Kallioniemi and Kauhealampi, because the late, "high-T" cordierite-sillimanite assemblage (cf. Hiltuspuro) is absent (Appendix 2). Decomposition of primary biotite and plagioclase into chlorite and muscovite has proceeded with increasing deformation from the margins towards the centre of zones. The central part of the zone has, however, been neomineralized to coarse-grained kyanite rock with unoriented kyanite laths, up to 10-15 cm in diameter, caused by static mineral growth. Kyanite replaces mainly muscovite. A sequence of metasomatic zones from tonalitic Pl-Qtz-Bt to Chl-Ky-Ms-Qtz±Pl was formed. Segregation of mica- and quartz±plagioclase-rich domains is common.

At *Turkkivaara* (local boulders only) relics of early shearing structures, e.g. bending of Archaean



Fig. 8. Kyanite-staurolite paragenesis surrounded by later cordierite and sillimanite in a narrow D_{P2} zone at Hiltuspuro. Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.

foliation, are well preserved (Fig. 9). The oldest product of shearing and concomitant metasomatism is a foliated biotite-plagioclase rock – kidney-shaped plagioclase porphyroblasts in biotite mass – overgrown by static kyanite and staurolite. The kyanite is rimmed by later and alusite, which also occurs as large subidioblastic porphyroblasts. Cordierite and scarce fibrous sillimanite are the most recent phases. Cordierite rims and corrodes kyanite and andalusite (Fig. 10), and has locally altered into muscovite and chlorite.

The mineral growth succession in the kyanite rock about one kilometre east of Havukkalammit, *east Havukkalammit*, is much the same as that at Turkkivaara. The relation of the rock to its surroundings is unknown. Andalusite, which rims kyanite and is itself surrounded by cordierite, is xenoblastic and corroded, but exists in equilibrium with cordierite. There is metasomatic segregation of nearly monomineralic muscovite and biotite rocks. Micas have been replaced isovolumetrically by cordierite (Fig. 11).

At Korpilampi, Havukkalammit and Levävaara the relation between the coarse-grained, weakly oriented kyanite rocks and shearing is not as clear as at the westernmost sites described above. The kyanite rocks have irregular, sharp or gradual, or breccia-like replacement contacts with the variably altered tonalitic country rocks. The breccia fragments are strongly assimilated and rather mobile in appearance, suggesting brecciation in a ductile state (Fig. 12). At Levävaara, a Palaeoproterozoic metadiabase dyke shows well-developed amphibolite-facies mineral schistosity, and the country-rock tonalite is blastomylonitic-mylonite gneissic in fabric. The mica-rich kyanite rock is undeformed; it has only local late crenulations (cf. Hiltuspuro), due to which the large kyanite crystals were broken and altered into muscovite. Metadiabase shows alteration that occurred after the formation of the peak metamorphic assemblages in its contact zone, indicating that the alteration process postdated the amphibolite facies parageneses and deformation of metadiabase.



Fig. 9. Metasomatic kyanite-andalusite-cordierite rock penetrating tonalite (in upper right corner) at Turkkivaara. The foliation in tonalite is bent in the direction of early shearing, and the foliation is still visible in the zone itself. The early biotite-plagioclase paragenesis (Bt = black and Pl = white, in lower left corner) covered by static mineral growth of cordierite (grey, biotite foliation as relic) and Al-silicates (white prisms in middle of photograph). Photo by M. Pajunen.



Fig. 10. Kyanite rimmed by later and alusite and cordierite at Turkkivaara. Field of view c. 4 mm wide. Crossed nicols. Photo by M. Pajunen.



Fig. 11. Foliated, monomineralic biotite rock (cf. Fig. 9) overgrown by cordierite at east Havukkalammit. Biotite foliation is preserved as relic in internal structure of cordierite. In the middle of the cordierite porphyroblast new, more randomly oriented muscovite has crystallized instead of biotite. Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.



Fig. 12. Mobile, brecciated replacement contact of coarse-grained kyanite rock and tonalite at Levävaara. Metasomatic zoning, for example, a zone rich in tourmaline (black patches in upper-left corner), is well developed. Photo by M. Pajunen.



Fig. 13. Metasomatic biotite-plagioclase (black), plagioclase-magnetite-biotite (white) and epidote-clinoamphibole (darkish grey) zones at Havukkalammit. Photo by M. Pajunen.

Compositional variation is pronounced. Extremely well-developed metasomatic zoning exists at Havukkalammit, where zones from the tonalitic assemblage, Pl-Qtz-Bt±Hbl, to quartz-free, medium-grained Pl-Mag-Bt and Ep-Cam (Fig. 13) and also to coarse-grained Bt-Pl, Bt-Ky-Pl and Bt-Ky assemblages were formed. At Havukkalammit and Korpilampi the late "high-T" cordierite-sillimanite assemblages are absent, and the kyanitebiotite pair is stable, although fresh, exceptionally anorthite-rich and irregularly, even oscillatory, zoned plagioclase rims (see later) exist between biotite and kyanite. Kyanite occurs as fresh and uncorroded laths, up to 50 cm long, which in places are surrounded by fibrous kyanite of a later generation (Fig. 14). At Levävaara a late cordierite-sillimanite assemblage, albeit weaker than at Hiltuspuro, developed. The texture of kyanite and andalusite indicates that they crystallized more or less concurrently, and that there is a progression from chlorite-muscovite to kyanite-staurolite (Fig. 15).

At Varpuniemi and Matovaara (Appendix 2) a fully recrystallized granoblastic fabric formed in northwest-southeast-trending shear zones crosscutting the Archaean tonalite. At Varpuniemi, staurolite grains, about 1 cm in size, and rare kyanite porphyroblasts occur in a quartz-free assemblage, and at Matovaara there is an exceptional orthoamphibole-bearing rock with unoriented amphibole laths, 2–4 cm long.

At *Tuomaanvaara*, irregular cordierite-chlorite rock patches, corresponding to the cordierite-rich patches at Hiltuspuro, overprint the Archaean tonalite without distinct deformation (Fig. 16). The early Chl-Ms-Qtz and Ky/And-St assemblages (see above) are absent.

The kyanite rock at *Teljo* (one outcrop) is fully comparable to the narrow zones and cordieriterich patches at Hiltuspuro. Locally, cordierite grains replace plagioclase along grain boundaries (Fig. 17).

The kyanite rock at *Tetrilampi* differs from that at Hiltuspuro in its more intense sillimanitization;



Fig. 14. Coarse-grained kyanite (Ky_1) and fibrous kyanite (Ky_2) generations at Havukkalammit. Fibrous kyanite phase was identified by X-ray methods, and it has inclined extinction along <u>c</u>-axis. Field of view c. 2 mm wide. Plane polarized light. Photo by M. Pajunen.



Fig. 15. Kyanite-staurolite paragenesis overgrowing coarse-grained chlorite-muscovite (Ms_1) paragenesis at Levävaara. Retrograde fine-grained muscovite (Ms_2) rimming kyanite. Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.



Fig. 16. Cordierite-rich patch, surrounded by chlorite-rich zone, has grown isovolumetrically on foliated and folded Archaean structure of gneissic tonalite at Tuomaanvaara. Photo by M. Pajunen.



Fig. 17. Cordierite replacing plagioclase along grain boundaries at Teljo. The replacement is locally total, and only small relics of previous phases are preserved (cf. Fig 7). Field of view is c. 4 mm wide. Crossed nicols. Photo by M. Pajunen.

sillimanite even forms narrow monomineralic zones overgrowing kyanite. Some kyanite grains follow the grain boundaries of earlier plagioclase, indicating alteration along grain boundaries in the same manner as the cordierite at Teljo. Muscovite is a significant by-product of cordierite after kyanite and biotite.

Notes on accessory minerals of kyanite rocks

Some features regarding accessory phases need to be emphasized (see Appendix 2). Abundant rutile crystallized in chloritization and/or sericitization of biotite at most of the sites and was stable during metasomatism. Ilmenite is scarce in stauroliteand orthoamphibole-bearing assemblages. Apatite occurs regularly and abundantly, especially in altered rocks rich in cordierite, indicating phosphorus mobility during metasomatism. Fresh xenotime inclusions in cordierite in the southern occurrences imply phosphorus enrichment and rearrangement of radioactive components and yttrium (see section "Isotope geology", pp. 110-112). REE mobility during the alteration is shown by local allanite crystallization. Radioactive minerals in chlorite-rich assemblages, not affected by the latestage event, are often metamictic. Local enrichment of boron is manifested in crystallization of tourmaline porphyroblasts, up to 10 cm in diameter, at Levävaara (Fig. 12).

MINERAL CHEMISTRY

The chemical compositions of minerals were analysed on polished thin sections at the Geological Survey of Finland in Espoo using the wave-length dispersive technique and a Jeol JCXA – 733 Superprobe. The acceleration voltage was 15 kV. The electron beam current was 15 nA for cordierite, plagioclase and muscovite and 25 nA for other phases; the beam diameter was 1 μ m for garnet and 10 μ m for other minerals. Natural mineral standards were employed for all elements except Al, for which synthetic corundum was used. ZAF procedures were applied to correct the analytical data. The results confirmed by several analyses are shown as averages of *n* analyses in Appendices 3–10. Ring number refers to the local mineral assemblage analysed within each thin section.

Kyanite

Kyanite is, in general, pure Al₂SiO₅, but minor impurities

of iron (0.45–0.54 wt% FeO) were found in the kyanite from a narrow zone at Hiltuspuro. The coarse-grained kyanite from Havukkalammit contains small amounts of iron (0.05–0.51 wt% FeO; analyses 1–3 on p. 13 in Horneman & Hyvärinen 1987).

Cordierite

Cordierite is fresh, as only weak pinitization along grain boundaries and fractures is sometimes observed. The cordierites $[(Mg,Fe)_2^M(Al,Fe)_4^{T1}(Si,Al)_5^{T2}O_{18}]$ analysed from Hiltuspuro and Tetrilampi vary slightly in composition (Appendix 3, analyses 1–14). That at Tetrilampi is richest in Mg: $X_{Mg}^{crd} = Mg/(Fe+Mn+Mg)$ is 0.84–0.85 (analyses 13– 14). At Hiltuspuro, the highest $X_{Mg}^{crd} = 0.84$ was analysed from a coarse-grained kyanite-chlorite-cordierite rock (analyses 11–12). In orthoamphibole-bearing assemblages (analyses 1–7), X_{Mg}^{crd} varies in the range 0.81–0.83 and the cordierite richest in iron, $X_{Mg}^{Crd} = 0.77-0.79$, is in staurolite-bearing assemblages (analyses 8–10). The compositional zoning in cordierite is small. Trace element Na₂O has a slight variation; the coarse-grained kyanite-chloritecordierite rocks have the highest values.

Staurolite

The composition of staurolite $[(Fe,Mg)_2^{M4+M6}(A1,Fe)_9^{M6}(Si,A1)_4^TO_{23}(OH)]$ with yellow pleochroism varies very little in samples analysed from the narrow alteration zones of Hiltuspuro. The $X_{Mg}^{st} = Mg/(Fe+Mn+Mg)$ values range from 0.26 to 0.28 (Appendix 4, analyses 15–18). Analysis 15, which has the lowest X_{Mg}^{st} is of a small, younger idioblastic grain in cordierite, and analysis 16 is of a corroded, xenoblastic grain in the same paragenesis.

Garnet

The garnet-quartz rock at Hiltuspuro has various garnet [(Ca,Fe,Mg,Mn)₃(Al,Cr)₂(Si,Ti)₃O₁₂] parageneses, even in one sample: (1) Grt-Bt-Chl-Qtz (Appendix 5, analyses 19–22) and (2) Grt-Pl-Ms-Qtz (analyses 23–25). The garnet is poor in Mg, X_{Prp} ranging from 0.04 to 0.09. Paragenesis (1) shows two generations of garnet: the older core ($X_{Alm} = 0.60$ and $X_{sps} = 0.21$, analysis 20), which has altered into chlorite, and the younger rim generation ($X_{Alm} = 0.60-0.62$ and $X_{sps} = 0.22-0.23$, analyses 19 and 21–22) enveloping chlorite. X_{Grs} is low (0.09–0.11). Syntectonic garnet in the muscovite-bearing paragenesis (2) (analyses 23–25, a grain about 6 mm in diameter), with weak zoning in composition from core to rim, has X_{Alm} ranging from 0.77 to 0.82. Mn and Ca contents are lower than in paragenesis (1): $X_{sps} = 0.10-0.13$ and $X_{Grs} = 0.03-0.04$.

Orthoamphibole

Orthoamphibole was found in one outcrop at Hiltuspuro and at Matovaara (not analysed). Complex substitutions and the impossibility of analysing Fe^{2+} and Fe^{3+} separately by microprobe complicates determination of the amphibole formula [(Vac,Na,K)^A(Na,K,Ca,Mn,Mg,Fe)^{M4} (Fe,Mg,Mn, $Ca)_{2}^{M1+M3}(Al,Ti,Fe,Mg)_{2}^{M2}(Si)_{4}^{T1}(Si,Al,Ti)_{4}^{T2}O_{22}(OH)_{2}]$ (Vac = vacancy) (Robinson et al. 1982). The mineral formulae were normalized to 15 cations and all Fe was considered as Fe²⁺. Mg and Fe were divided between M4, M1+M3 and M2 sites according to the molecular portion, $X_{Mg}^{Oam} = Mg/$ (Fe+Mg) (Appendix 6, analyses 26-34). According to the classification of Leake (1978), all the orthoamphiboles are anthophyllitic except that of analysis 27, which is gedritic in composition. As products of the same chlorite breakdown reaction, coexisting gedrite and anthophyllite (analyses 27 and 28) occur as separate grains in the paragenesis, Ged-Ath-Crd-Chl-Qtz. Gedrite has lower $X_{Mg} = 0.61$ and higher Na = 0.43 p.f.u. (p.f.u. = atoms per formula unit) than coexisting anthophyllite, 0.65 and 0.07, respectively. Altot in gedrite is 2.28 p.f.u. as against 0.55 p.f.u. in coexisting anthophyllite. The substitutions of Na and Al in gedrite are due to edenitic (Vac,Si=Na,Al^{IV}) and tschermakitic (Mg^{VI},Si^{IV}=Al^{VI},Al^{IV}) substitutions (cf. Robinson et al. 1982). The other orthoamphiboles of the same rock are anthophyllites, with X_{Mg} between 0.67 and 0.69 and Al^{tot} ranging from 0.62 p.f.u. in the complex assemblage, Ath-Crd-Pl-Bt-Chl-Qtz-Ky (analysis 26), to 0.24-0.44 in the assemblage, Ath-Crd-Qtz±Pl (analyses 32-34). Anthophyllites 29–30 with $X_{Mg} = 0.68$ and $Al^{tot} = 0.47-0.39$ replace biotite in the paragenesis, Ath-Bt-Crd.

Plagioclase

Plagioclase is albitic, $X_{Ab} = 0.90-0.98$, in kyanite-chloritecordierite rocks at Hiltuspuro (analyses 41–43 and 45–47) and Tetrilampi (analysis 50) (Appendix 7). It is weakly zonal. The kyanite rock at Havukkalammit (analyses 35– 40) is exceptional in having plagioclase rich in anorthite component, ranging from 0.57 to 0.86 (analyses 35–37) in older grains rich in inclusions and from 0.59 to 0.75 (analyses 38–40) in plagioclase reaction rims between kyanite and biotite. Its heterogeneous optical extinction properties indicate irregular variation in composition. The plagioclase of the garnet-quartz rock (analyses 48–49) at Hiltuspuro generally contains somewhat more anorthite component than the other rocks of the area. The potassium content is low in all samples.

Biotite

Biotites $[(Vac, K, Na)^{A}(Mg, Fe, Ca)^{M1}(Mg, Fe, Mn, Ti, A1)_{2}^{M2}(Si)_{2}^{T1}(Si, A1)_{2}^{T2}O_{10}(OH)_{2})]$ at Havukkalammit, Hiltuspuro and Tetrilampi (Appendix 8) are intermediate in Al-Si/oc-tahedral cation substitution. At Hiltuspuro, the staurolite-bearing parageneses have lower $X_{Mg}^{Bt} = Mg/(Mg+Fe)$ values, 0.60–0.64 (analyses 70–74), than the other altered parageneses. X_{Mg}^{Bt} is 0.71–0.73 (analyses 63–68) in orthoam-phibole-bearing parageneses, 0.73–0.74 (analyses 75–76) in the coarse-grained assemblage, Bt-Chl-Crd-Chl-Pl-Qtz, and high, 0.76 (analysis 77), in Bt-Chl-Crd aggregates. The X_{Mg}^{Bt} values of the biotite in the narrow zones are higher

(0.72–0.75) at Tetrilampi (analyses 59–62) than in corresponding zones at Hiltuspuro. In the coarse-grained Bt-Ky-Pl paragenesis at Havukkalammit, X_{Mg}^{Br} ranges from 0.67 to 0.72 (analyses 51–58) and is low, 0.32–0.36 (analyses 78–79), in the garnet-quartz rock at Hiltuspuro. The Ti content is evidently buffered by rutile to 0.05–0.07 p.f.u. except at Havukkalammit, where it is 0.03–0.05 p.f.u (Fig. 18).

Chlorite

Chlorite [(Mg,Fe,Mn)⁴¹₄(Al,Mg,Fe,Mn)²²₂(Si,Al)²⁷₂(Si)⁷¹₂O₁₀ (OH)₈] (Appendix 9) exists in various assemblages, but its composition varies only slightly. At Hiltuspuro, the X^{Chl}_{Mg} = Mg/(Fe+Mn+Mg) values are high, from 0.75 to 0.80 in altered rocks (analyses 83–90), being highest in the Chl-Bt-Crd aggregate (analysis 90), where biotite, too, is richest in Mg. At Havukkalammit, X^{Chl}_{Mg} is between 0.70 and 0.75 (analyses 84–86). In the garnet-quartz rock, X^{Chl}_{Mg} is 0.34–0.36 (analyses 91–93).

White micas

Early muscovite $[(Vac, K, Na)^A(Mg, Fe^{2+}, Mn_x^{M1}(A1, Fe^{3+}, Ti)_{2x}^{M2}(A1_{1-x}, Si_{3+x})_4^{T1+T2}O_{10}(OH)_2]$ with chlorite occurs as inclusions in kyanite at Havukkalammit (analyses 94–95) (Appendix 10). At Tetrilampi the muscovite is in the paragenesis, Ms-Crd-Ky/Sil-Bt-Qtz±Pl (analyses 99–103). $X_{Na} = Na/(Na+K)$ is 0.18–0.19 at Havukkalammit and 0.25–0.29 at Tetrilampi. There is slight substitution of celadonite component in muscovite. The small mica grains at Hiltuspuro generated by the alteration of albitic plagioclase in the plagioclase-cordierite reaction rim are paragonitic in composition, with $X_{Na} = 0.85$ –0.88 (analyses 96–97) and a high Mg content, 0.14–0.58 p.f.u. In the garnet-quartz rock at Hiltuspuro X_{Na} is 0.13 (analysis 98).

FLUID INCLUSIONS

To shed light on fluid composition during metasomatism, a fluid inclusion study was made on carefully gathered samples. The likelihood of finding representative fluid inclusions was good, because the rocks are coarse grained, and generally well preserved from late deformations.

Fluid inclusions in 0.3-mm-thick, doubly-polished plates cut from kyanite-chlorite-cordierite rock (from the outcrop shown in Figs. 2b, 3 and 5) from southeast Hiltuspuro were investigated using a Linkam THMSG 600 programmable heating/freezing stage attached to a Leitz Ortholux Pol transmitted light microscope (see Shepherd et al. 1985). The stage was calibrated with a set of syn-



Fig. 18. Biotite compositions in kyanite rocks and pre-metasomatic Archaean tonalites on TiO_2 vs. X_{Mg} diagram.

thetic fluid inclusion standards (see Sterner & Bodnar 1984). At a temperature interval of -60° to 300°C, the recorded temperatures of phase transitions have a precision of $\pm 0.2^{\circ}$ to 2.0°C. Analytical errors are thus insignificant in terms of geological interpretation.

The data obtained from 60 carefully selected fluid inclusions (c. 5 to 70 μ m in the longest dimension) in kyanite and cordierite are summarized in Table 1. No fluid inclusions were found in andalusite from Levävaara. Two types of fluid inclusion were identified: (1) aqueous, two-phase inclusions with moderate salinity and (2) monophase, apparently pure liquid CO₂ inclusions without a visible amount of water. The salinities and densities of the aqueous and carbonic inclusions were calculated from the last ice melting temperatures and homogenization temperatures to liquid and were modelled in binary (H₂O-NaCl) and unary (CO₂) systems in the equations of Bottinga and Richet (1981) and Brown and Lamb (1989), respectively, using the FLINCOR program (Brown 1989). The isochore slopes presented in Fig. 31 were calculated using the above equation of states and FLINCOR. The relevant data for constructing isochores are listed in Table 1.

The fluid inclusions were further subdivided into primary, pseudosecondary and secondary inclusions (Table 1) using criteria given by Roedder (1984). Their mode of occurrence is shown in Fig. 19. In kyanite, primary aqueous inclusions are tubular in shape and aligned parallel to the c-axis of the crystal (Figs. 19 and 20). In cordierite, pseudosecondary carbonic inclusions occur in small groups and their shape is usually of negative crystals. Secondary carbonic inclusions in cordierite, and aqueous inclusions in kyanite and cordierite occur in healed fractures with rounded to irregular shapes. The mutual cross-cutting features of these secondary inclusion trails in cordierite suggest that the carbonic inclusions predate the aqueous inclusions. It is assumed that the

Inclusion type/mineral	$T_{h}H_{2}O$ T°C	T _m H ₂ O T°C	Salinity eq.wt % NaCl	$T_h CO_2 T^\circ C$	T _m CO ₂ T°C	Density g/cm ³	Paragenetic classification
H ₂ O/kyanite	251 to 264	-9.5 to -11	13 to 15	-	-	0.92	primary
H ₂ O/kyanite and cordierite	227 to 245	-7.5 to -13	11 to 17	_	_	0.93 to 0.95	secondary
CO ₂ /cordierite pseudosecondary	-	-	-	8.5 to 13.5	-56.6	0.83 to 0.87	
CO ₂ /cordierite	-	-	-	21 to 29	-56.6	0.63 to 0.7	secondary

Table 1. Summary of fluid types, microthermometric data and paragenetic classification of fluid inclusions in kyanite and cordierite of kyanite-chlorite-cordierite rock at Hiltuspuro (x=7070.51, y=4485.25).

 T_{h} = temperature of total homogenization of inclusion contents into liquid

 T_m = melting temperature of ice and carbon dioxide



Fig. 19. Schematic illustration of mode of occurrence of primary (P), pseudosecondary (PS) and secondary (S) fluid inclusions in cordierite and kyanite.

primary and pseudosecondary inclusions provide information on fluids passing through the rock during crystallization of the mineral involved.

METASOMATIC-METAMORPHIC REACTION HISTORY

Metasomatism, the mass transfer and reorganization of rock-forming components, occurs whenever a moving fluid is out of equilibrium with the local rock (Korzhinskii 1970, Thompson 1988). The process is maintained by infiltration and/or diffusion processes in solid rock (e.g. Korzhinskii 1970, McCaigh & Knipe 1990). Both processes have a tendency to develop monomineralic metasomatic zoning. In infiltration metasomatism, components are transported by migrating pore solutions, and the zones that are formed have homogeneous phase compositions. In diffusion metasomatism, in contrast, the components migrate by diffusion through pore solution film, and phase



Fig. 20. Primary H_2O inclusion in kyanite at Hiltuspuro (x=7070.51, y=4485.25) from the outcrop shown in Figs. 2b, 3 and 5. Photo by M. Poutiainen.

compositions may vary within zones. Metasomatic zones formed simultaneously have sharp boundaries in which one component becomes perfectly mobile and the number of inert components decreases by one, one phase then being totally replaced by a new one. Thus, the fluid composition, especially its pH-Eh states and the character of dissolved components, vary from one zone to another. Diffusion may change the boundaries, making them gradual. Generally, both processes act concurrently in metasomatism. In nature, metasomatic zoning is seldom systematic, and the zones constitute a complex mixture (Korzhinskii 1970).

The metasomatic-metamorphic reaction history of the kyanite rocks can be established by combining the information available on their common features. The metasomatic-metamorphic succession can be divided into early, late and cooling stages. In the westernmost targets, the earliest stage was connected with shearing (Fig. 9). This was overprinted by static-state metasomatism in the centre of the zone, which was the part most susceptible to fluid flow. Further east, the fluid action of the early stage gave rise to mobile breccia-like replacement structures (Fig. 12) and welldeveloped metasomatic zoning (Fig. 13). During the late stage, fluid-induced metasomatism occurred along semi-brittle fractures and mineral grain boundaries under unstrained conditions (Figs. 4 and 17). The cooling stage was associated with weak fracturing and static retrogression.

Early stage

Metasomatic zoning and mineral assemblages

Early parageneses or their relics are preserved in the coarse-grained kyanite rocks at Hiltuspuro and to the northwest of it, but in the south they are more or less overprinted by later processes or, as at Tuomaanvaara, were not formed at all. Characteristic mineral growth successions and metasomatic zones during progressive metasomatism are compiled in Table 2.

Large changes in volume, even losses of up to 60%, have been described from shear zones (e.g. Selverstone et al. 1991). The estimation of volume changes and the amount of chemical mass trans-

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Occurrence	Pre-metas	omatic stage	Metasomatic stage	es				Cooling stage
(in Fig. 1)	Primary ass	emblages	Early-stage mineral	growth		Late-stage mine	ral growth	
	Rock type	Original assemblage	Initial assemblages and metasomatic zones	Successive mineral growth	Transitional or unknown stage	Assemblage in alteration front	Final	Retrograde alteration
Tuomaanvaara (1)	tonalite	Pl-Bt-Qtz	not developed			Chl-Qtz and Bt in cordieritization	Crd	pinitization
Turkkivaara (2)	tonalite	Pl-Bt-Qtz	Bt-Pl±Ms1 Chl1-Ms Bt	Ky±St	And	•	Crd-Ms Crd ± Sil	Ms2 Chl2
Kallioniemi (3)	tonalite	PI-Bt-Qtz	Ms-Chl-Pl-Qtz	Kv			not developed	
Varisniemi (4)	tonalite	PI-Bt-Qtz	Chl-Pl	St±Kv			not developed	Ms
Korpilampi (5)	tonalite	PI-Bt-Qtz	Bt-Pl-Qtz-Chl ± Ms1	Ky			not developed	Ms2
Kauhealampi (6)	tonalite	PI-Bt-Qtz	Chl-Qtz ± Ms1 Chl-Pl	Ку			not developed	Ms2
Havukkalammit (7)	mit (7) tonalite PI-Bt-Qtz mmit (7) not known not known		Ms-Chl-bearing assemblages	Ky1-Bt±Pl1(rich in An) Bt-Chl±Pl Ep-Pl±Cam Pl-Mag-Bt	Ky2 PI2 (rich in An)			
E-Havukkalammit (7)	not known	not known	Bt-Pl Bt-Qtz Ms	Ку	And	•	Crd	
Matovaara (8)	altered tonalite	PI-Bt-Qtz	Bt-PI-Qtz Qtz (veins)		4		Oam	Chl
Levävaara (9)	tonalite	Pl-Bt-Qtz	ChI-PI-Bt ChI-PI-Qtz ChI-PI ChI-Ms1 ChI-Bt Bt-PI ChI-Bt-Sph PI-ChI-Tou Qtz Bt	Ky±St Ky-And And Crn-Ky Crn			Crd-Sil	Ms2
miituspuro (10)	tonalite	PI-Bt-Qtz	Chi1-PI-Bt Chi1-Bt Bt Bt-PI Qtz (veins)	Ky-St		Chl in cordieritization	Crd-Sil Crd-Oam Crd-Qtz Crd	Ms Chl2 carbonatization
Teljo (11)	tonalite	PI-Bt-Qtz		Ky-Bt-Pl relics in Crd		······	Crd ± Sil	Ms
						Chl (*) see above	•	pinitization
Egyptinkorpi (12)	not known	not known	Bt-Ms1 relics in Ky	PI-Qtz-Ky		•	Sil	Ms2
Tetrilampi (14)	tonalite and migmatite	PI-Bt-Qtz		Ку		►	Crd-Ms1 ± Sil Crd Sil (veins)	Ms2 Chl



Fig. 21. Compositions of early-stage phases and pre-metasomatic phases in tonalites shown on Si/(Si+Al) vs. Mg/(Mg+Fe+Mn) diagram. The mineral compositions are from Havukkalammit; the compositions of staurolite refer to Hiltuspuro. The directions of metasomatism are shown schematically by arrows and the general trend of alteration by hatched arrow. Typical mineral assemblages are indicated by numbers in parentheses.

fer (cf. Gresens 1967, Grant 1986) during metasomatism is beyond the scope of our study. Thus, only relative changes in composition are shown in the Si/(Si+Al) vs. Mg/(Mg+Fe+Mn) diagram (Fig. 21). The representative mineral assemblages are indicated by the numbers in parentheses. The mineral compositions from Havukkalammit are used; the compositions of staurolite refer to Hiltuspuro, and those of pre-alteration phases in amphibolite facies tonalites are given as "precursor". The early-stage mineral compositions were clearly affected by the later "high-T" conditions, the exchange reactions between phases in particular, and by the late-stage fluid (probably causing the ferromagnesian phases to be enriched in Mg), even though the primary mineral assemblages may have been preserved. Not all the important changes caused by leaching or redeposition during metasomatism can be shown in the diagram. The changes in alkali and calcium contents are characterized only by the line bordering the field of Ca-Na-K-free phases, which descriptively emphasizes the tendency of the system to form Ca-alkali free assemblages in the most altered zones. It is not possible, for instance, to show the accumulation of calcium-rich phases at Havukkalammit.

In the northwestern targets, that is, Kallioniemi and Kauhealampi, successive metasomatism accompanied by shearing caused progressive metasomatic zoning towards the centre of the zones. Alteration started with the disappearance of biotite and a decrease in plagioclase content with a concomitant increase in chlorite and muscovite contents. These changes indicate sodium and calcium leaching with a relative increase in Al/(Al+Si) and ferromagnesian components in rock. During advanced metasomatism, when potassium and silica activities diminished and quartz was fully leached out, kyanite and local staurolite were stabilized, replacing muscovite and chlorite. Metasomatism resulted in the compositions characterized by phase fields (4) and (5) in Fig. 21.

Further east, early-stage metasomatic segregation was more extensive and complex metasomatic zoning (Table 2) developed, as illustrated by the arrows in Fig. 21. The early-stage metasomatic zones with gradual or sharp boundaries can be detected, but at Hiltuspuro and further south they have often been modified, or even destroyed, by later mineral growth. In the southernmost occurrences it is impossible to separate early- and latestage mineral growths from each other with the aid of microstructures. At Levävaara and Havukkalammit, well-developed and sharply-defined metasomatic zones (Fig. 13) show diverse complex leaching and deposition of components.

The general alteration trend exhibited by the shaded arrow in Fig. 21 led to the disappearance of quartz; locally quartz is concentrated, redeposited, in fractures. An extreme decrease in silica activity occurred at Levävaara, where corundum stabilized with chlorite and aluminium silicates. The decrease in alkalies and Ca and the relative increase in aluminium and Fe-Mg were important processes that changed the mineral compositions towards parageneses rich in chlorite and kyanite/ andalusite±staurolite and poor in plagioclase and biotite as silica activity declined. The relative decrease in iron is also shown by the high X_{Mg} of biotite and chlorite compared with that of the less magnesian primary biotite and hornblende, but this change may partly be a result of later fluid activity. Examples of the deposition of leached components are seen in several occurrences. At Havukkalammit the exceptionally high calcium content, e.g in Ep-Pl-Cam paragenesis, is due to extensive calcium redeposition. Similarly, the patches of monomineralic biotite or chlorite are evidence of the accumulation of components during metasomatism. In the metasomatic zones in kyanite rocks one phase is often enriched at the cost of the others, thus promoting the formation of monomineralic rocks.

The zones with large tournaline crystals at Levävaara (Fig. 12) are examples of the deposition of trace elements such as boron.

Mineral reactions

The early-stage reactions were catalysed by saline H_2O fluid as indicated by the fluid inclusions in kyanite (Table 1). Because the fluid inclusion data permit a rough estimation of fluid composition, reactions can be delineated only in simplified forms.

During early-stage metasomatism, silica was a mobile, dominantly consumed component. Above 200°C, silica is usually released into solution and it combines with water molecules to form a notional H₄SiO⁰₄ aqueous complex. The dissolution of silica is independent of the concentration of ligands such as chloride or of pH. Under alkaline conditions, silica may dissolve and form alkali silica complexes such as KH₃SiO₄⁰; silica dissolution then becomes pH dependent. However, the presence of small amounts of chlorine is sufficient to prevent high pH. The solubility of SiO₂ in NaCl solutions increases relative to that in pure water (Sorokin & Dadze 1995, Yardley & Shmulovich 1995). High salinity of fluid during the early stage is indicated by fluid inclusions. Thus, non-alkaline properties during silica leaching are postulated for kyanite rocks. Acid properties of the fluid are also obvious from the mineral assemblages rich in basic components and poor in silica (Korzhinskii 1964). With the exception of reactions producing quartz, silica is not considered in the following formulations.

Early-stage shearing was accompanied by decomposition of country rock biotite to chlorite and rutile in the reaction (observed reactants in **bold**)

(1) $Fe-Mg-Ti-Bt + H^+ \rightarrow Mg-Chl+Rt+Qtz+ K^++Fe^{2+}$,

and formation of muscovite, which, in places, overgrows chlorite. Plagioclase (oligoclase-andesine) has been sericitized; thus, it is likely that the K^+ released in reaction (1) was fixed in muscovite and that especially Na₂O was leached out of the system in the reaction

(2) $\mathbf{Pl}+\mathbf{K}^+ \rightarrow \mathbf{Ms} + \mathbf{Qtz} + \mathbf{Na}^+ + \mathbf{Ca}^{2+}$.

This process is supported by the increasing quartz content towards the muscovitized part of the shear zones. Locally, white mica (not analysed) was also produced by leaching of ferromagnesian components out of biotite, e.g. in the reaction

(3) **Bt**+H⁺ \rightarrow white mica + Rt+Fe²⁺+Mg²⁺,



Fig. 22. Leaching of biotite to white mica (seen in upper-right corner) in kyanite-staurolite contact at Levävaara. Later muscovite rims kyanite and staurolite. Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.

as shown by a sharp alteration front in biotite (Fig. 22).

The microstructures indicate that kyanite and andalusite crystallized more or less simultaneously at Levävaara, whereas at Turkkivaara and east Havukkalammit andalusite clearly postdates kyanite. Andalusite preceded cordierite, but the exact timing of its crystallization with respect to stage cannot be established. Nevertheless, many reactions are shared by both kyanite and andalusite formation. Kyanite and andalusite with corroded plagioclase (oligoclase-andesine) inclusions (Fig. 23) and kyanite replacing plagioclase along grain boundaries refer to decomposition of plagioclase to aluminium silicate directly in the leaching reaction

(4) $\mathbf{Pl} + \mathbf{H}^+ \rightarrow \mathbf{And}/\mathbf{Ky} + \mathbf{Qtz} + \mathbf{Na}^+ + \mathbf{Ca}^{2+}$.

Andalusite-quartz symplectite in the andalusiteplagioclase interface and small quartz grains in the contact of kyanite with plagioclase are interpreted as products of such a reaction. SiO_2 often dissolved in fluid and deposited elsewhere in dilatational fractures. The kyanite and andalusite frequently contain corroded muscovite inclusions; in such cases Al-silicate formation is attributed to the decomposition of white mica, muscovite and/or paragonite, in reactions (2),

- (5) $\mathbf{Pl}_{(Ab-Olg)} + H_2O \rightarrow \mathbf{Pg} + \mathbf{Qtz} + Ca^{2+}$ and
- (6) $Ms/Pg+H^+ \rightarrow Ky/And+K^+/Na^+$.

Chlorite and muscovite occur as separate inclusions, often as homoaxial relics after larger grains, in kyanite indicating the leaching reaction

(7) **Chl+Ms**+H⁺
$$\rightarrow$$
 Ky+Fe²⁺+Mg²⁺+K⁺.

Silica and potassium dissolved in fluid, and/or a new generation of biotite formed, as exemplified by the biotite-kyanite rock at Havukkalammit. Biotite was also decomposed to kyanite in a



Fig. 23. Corroded plagioclase relics in metasomatic kyanite in narrow zone at Hiltuspuro. Kyanite rimmed by later cordierite, which also replaces biotite. Field of view c. 4 mm wide. Crossed nicols. Photo by M. Pajunen.

metasomatic reaction that released ferromagnesian components and potassium:

(8) $\mathbf{Bt} + \mathbf{H}^+ \rightarrow \mathbf{Ky} + \mathbf{Rt} + \mathbf{Fe}^{2+} + \mathbf{Mg}^{2+} + \mathbf{K}^+$.

At Levävaara, rare sulphurization of intermediate phases has led to corundum formation. Evidence of such a reaction,

(9) $Chl+Ms+S^{2+} \rightarrow Crn+Pyh+Py+Mg^{2+}+K^{+}+Si^{4+},$

is the sulphide inclusions in corundum. An example of corundum replacing white mica in the leaching reaction,

(10) $Ms+H^+ \rightarrow Crn+K^++Si^{4+}$

is shown in Fig. 24. The lack of sulphides or other indications of sulphurization reactions elsewhere in the kyanite rocks suggests their poor sulphideore potential. Aluminium is usually considered an immobile component in metamorphic processes but, as pointed out by Anderson and Burnham (1983), the solubility of aluminium depends on the prevailing conditions – especially on the (KCl, NaCl)/HCl ratio and, thus, on the pH of the fluid. Thus, the relative enrichment of kyanite rock in aluminium may be partly due to the precipitation of Al from solution (cf. Korzhinskii 1964).

Staurolite is a minor phase with kyanite but it is abundant in parageneses St-Chl-Pl and St-Chl-Ms-Pl. Staurolite is free of inclusions and the corroded grains are surrounded by later cordierite or by retrograde muscovite. In the light of the staurolite-bearing assemblages we suggest that staurolite was formed largely in prograde reactions between muscovite and chlorite, e.g.

(11) $Chl+Ms+H^+ \rightarrow St+Bt/(K^++Mg^{2+})$ and

(12) $\mathbf{Chl} + \mathbf{Ms} + \mathbf{H}^+ \rightarrow \mathbf{St} + \mathbf{Ky} + \mathbf{Mg}^{2+} + \mathbf{K}^+.$

Staurolite-forming reactions may also be writ-



Fig. 24. Corundum replaces muscovite at Levävaara. Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.

ten as isochemical ones. Staurolite, however, always occurs in quartz-free assemblages, which is not expected in the isochemical reaction after chlorite and muscovite. Thus, according to our interpretation, the more probable staurolite-forming reactions are the metasomatic leaching reactions described above.

It is also impossible to explain the large amounts of chlorite, biotite, plagioclase and kyanite present in certain rock compositions without complex fluid-induced leaching and deposition. The large, zonal albite porphyroblasts in biotite-plagioclase and chlorite-plagioclase rocks at Levävaara and Hiltuspuro record "episodic" redeposition of sodium. The rim of anorthite-rich plagioclase between biotite and kyanite at Havukkalammit (Fig. 25) is difficult to explain without metasomatic deposition of calcium from fluid to replace the ferromagnesian components and potassium in the kyanite-biotite interface. The reaction may be formulated as Aggregation of coarse-grained biotite and chlorite indicates deposition of potassium and magnesium. The compositions of biotite and chlorite may also have restabilized and homogenized (see Fig. 18 and Appendices 8 and 9) in an exchange reaction with the Mg-rich solution

(14) Fe-Bt+Mg²⁺ \rightarrow Mg-Bt+Fe²⁺+Rt.

As stated by Korzhinskii (1970), the metasomatic zones formed through the infiltration process have homogeneous phase compositions (see Fig. 18) and a tendency to form monomineralic zones (Fig. 11). Such features are well displayed in the far-advanced metasomatic zones of the early-stage kyanite rocks.

Late stage

Mineral assemblages

A sharp break between the early-stage and latestage mineral reactions is not indicated by the

(13) $\mathbf{Bt} + \mathbf{Ky} + \mathbf{Ca}^{2+} \rightarrow \mathbf{An}_{59-75} + \mathbf{K}^{+} + \mathbf{Mg}^{2+} + \mathbf{Fe}^{2+}$.



Fig. 25. Metasomatic plagioclase, An_{59-75} , reaction rim between kyanite and biotite at Havukkalammit. Field of view c. 2 mm wide. Plane polarized light. Photo by M. Pajunen.

mineral growth succession alone, because many of the mineral reactions described above continued during the late stage. Differences in structure are, however, striking. The typical late-stage cordierite-sillimanite-bearing parageneses crystallized in dilatational semi-brittle fracture zones (Fig. 4), in patches (Fig. 16) or along fractures in grain boundaries (Fig. 17). The pseudosecondary CO_2 fluid inclusions in cordierite indicate a change in fluid composition after the early stage.

Metasomatic zoning is not as pronounced as that of the early stage (Table 2). The mineral parageneses at Hiltuspuro and Tetrilampi are shown in the Si/(Si+Al) vs. Mg/(Mg+Fe+Mn) diagram (Fig. 26). The late-stage assemblages often occur in dilatational structures replacing earlier phases along their boundaries. The most characteristic feature of this stage is the crystallization of cordierite, which replaces all the other phases in the rock. Plagioclase and biotite are disappearing phases, and alkalies and calcium become mobile components. Biotite has often redeposited in the vicinity of the cordierite-rich portions of altered rocks. Late-stage metasomatism shifts the mineral compositions towards the Mg edge on the Mg-Fe composition plane, implying the Mg metasomatism was the main alteration process during the late stage as shown by the shaded arrow in Fig. 26. During the early stage, silica leaching was pronounced. In contrast, the late-stage assemblages rich in cordierite and quartz indicate higher silica activity.

Mineral reactions

Progression during the late stage is characterized by metasomatic replacement reactions, but isochemical reactions also occurred. Changes in crystallization conditions, especially in temperature, are evidenced by the polymorphic transformation

(15) $\mathbf{Ky} \rightarrow \mathbf{Sil}_{\mathbf{fibr.}}$ (Fig. 27).

Reaction (15), which is typical of the south of the area, and the corresponding kyanite-andalusite \pm sillimanite succession in the north of the area



Fig. 26. Compositions of late-stage phases in tonalites shown on Si/ (Si+Al) vs. Mg/(Mg+Fe+Mn) diagram. The mineral compositions are from Hiltuspuro and Tetrilampi. The directions of metasomatism are shown schematically by arrows and the general trend of alteration by hatched arrow. Typical mineral assemblages are indicated by numbers in parentheses.



Fig. 27. Kyanite has transformed to sillimanite at Hiltuspuro. Field of view c. 2 mm wide. Plane polarized light. Photo by M. Pajunen.

(cf. Turkkivaara and east Havukkalammit) apparently indicate small variations in local P-T paths (see later). Sillimanite is frequently a by-product of the cordierite-forming reactions. Staurolite and kyanite exist as corroded relics in cordierite (Fig. 8). Staurolite and quartz are not in contact with each other, suggesting the prograde, isochemical reaction

(16) $St+Qtz \rightarrow Crd+Sil$ (cf. Richardson 1968).

In our opinion, however, it is more plausible that staurolite decomposed in metasomatic reactions with Mg^{2+} in fluid. The reactions could be formulated as

(17) $\mathbf{St} \pm \mathbf{Qtz} + \mathbf{Mg}^{2+} \rightarrow \mathbf{Crd} \pm \mathbf{Sil}/\mathbf{Al}^{3+} + \mathbf{Fe}^{2+}$ or

(18) $\mathbf{Ky} + \mathbf{St} + \mathbf{Mg}^{2+} \rightarrow \mathbf{Crd} \pm \mathbf{Sil}/\mathbf{Al}^{3+} + \mathbf{Fe}^{2+}$,

both of which explain, in a simpler way, the observed amounts of phases and the mass balance of components, e.g. between magnesium and iron. Decomposition of plagioclase rarely produced fibrous sillimanite (cf. reactions 4–6). Sillimanite also formed in narrow shear zones by leaching of biotite and plagioclase (cf. reactions 4 and 8).

Cordierite has grown over all other rock-forming minerals, producing the final paragenesis, Crd-Qtz-Rt-Sil, with corroded relics of pre-existing phases (Fig. 7). Many cordierite-producing reactions are Mg metasomatic and need mobile components for stoichiometry. For example, replacement structures, in which cordierite exploits the earlier phases, such as country-rock plagioclase (Fig. 17), cannot be explained without total rearrangement of the elements, as in the reaction

(19)
$$Pl+Mg^{2+} = Crd+Qtz+Na^{+}+Ca^{2+}$$
 (Fig. 28).

A narrow rim of paragonitic white mica between the corroded plagioclase and cordierite shows that, at least locally, replacement of plagioclase by cordierite proceeded through a paragonitic mica phase formulated as



Fig. 28. Microstructure produced by metasomatic breakdown reaction of plagioclase with Mg-rich fluid to cordierite and quartz. Field of view c. 2 mm wide. Crossed nicols. Photo by M. Pajunen.

- (20) $\mathbf{Pl} + \mathbf{Mg}^{2+} + \mathbf{H}_2\mathbf{O} \rightarrow \mathbf{Pg}$ (with Cel component) and further
- (21) $\mathbf{Pg+Mg^{2+} \rightarrow Crd+Na^{+}}$.

Formation of cordierite through a mica phase is also postulated by the chloritic alteration rim in the cordieritization front, as exemplified at Tuomaanvaara (Fig. 16). A distinctive cordieriteforming reaction is the decomposition of biotite with kyanite in the reaction

(22)
$$\mathbf{Bt}+\mathbf{Ky}+\mathbf{Mg}^{2+} \rightarrow \mathbf{Crd}+\mathbf{K}^{+}+\mathbf{Fe}^{2+}\pm\mathbf{Rt}$$
 (Fig. 23).

Potassium was expelled from the system in all breakdown reactions of biotite and was deposited in the coarse-grained, randomly oriented biotite aggregates characteristic of the targets with abundant cordierite. Exceptionally, at Tetrilampi, muscovite crystallized in a cordierite-producing biotite-breakdown reaction, which may be formulated as isochemical

(23) $Bt+Ky/Sil \rightarrow Crd+Ms+Qtz$.

The isochemical chlorite breakdown reaction (24) was locally an important cordierite-forming reaction, too.

(24) $\mathbf{Ky} + \mathbf{Chl} + \mathbf{Qtz} \rightarrow \mathbf{Crd} + \mathbf{H}_2\mathbf{O}$

At Hiltuspuro, the orthoamphibole-cordierite assemblage replaces biotite and chlorite. The prograde, possibly isochemical, decomposition of quartz and chlorite produced orthoamphibole in the reaction

(25) **Chl+Qtz** \rightarrow **Ath**+H₂O,

and locally also cordierite in the reaction

(26) $\mathbf{Chl} + \mathbf{Qtz} \rightarrow \mathbf{Ged} + \mathbf{Ath} + \mathbf{Crd} + \mathbf{H}_2\mathbf{O}.$

Orthoamphibole is anthophyllitic in composition except in reaction (26), which also produced concurrent gedrite in fibrous aggregates (Fig. 29). Anthophyllite replaces biotite in the metasomatic reaction (27) $\mathbf{Bt}+\mathbf{Mg}^{2+} \rightarrow \mathbf{Ath}+\mathbf{K}^{+}+\mathbf{Fe}^{2+}$ (Fig. 30).

Note that cordierite- and anthophyllite-bearing rock is a real metasomatic rock with a tonalitic precursor preserved as a relic in outcrops. A similar example of magnesia metasomatism of metadiabase to cordierite-anthophyllite rock has been described from Hirvas, northern Finland, by Härme and Perttunen (1971).

Cooling stage

In general, progressive metasomatic-metamorphic assemblages undergo only weak alterations in late retrogression. Reactions such as

(28) **Ky/And** \rightarrow **Ms** (e.g. Fig. 15),

(29) $\mathbf{Crd} \rightarrow \mathbf{Chl}$ and/or \mathbf{Ms} ,

(30) $\mathbf{Bt} \to \mathbf{Chl}$ and

(31) **Oam** \rightarrow **Chl+Qtz** or **Crb**

are related to late hydration and carbonization processes during the cooling stage. Weak retrogression is a strong indication of quick termination of fluidization and rapid cooling. Secondary fluid inclusions in kyanite and cordierite record late mobility of both CO₂- and H₂O-bearing fluids. Sericitization also indicates an increase in the activity of potassium in fluid during retrogression.

METAMORPHIC HISTORY

Regional metamorphism

The main features of the Archaean metamorphic evolution were described in section "Geological outline" (pp. 76–77). To summarize, the regional metamorphic peak conditions of amphibolite to granulite facies were attained in the TTG areas, but only greenschist to amphibolite facies conditions in the greenstone belts. The peak metamorphic parageneses are overprinted by epidote-amphibolite or greenschist facies retrogression. Ac-



Fig. 29. Chlorite-quartz breakdown reaction has produced anthophyllite, gedrite and cordierite at Hiltuspuro. Field of view c. 2 mm wide. Crossed nicols. Photo by M. Pajunen.



Fig. 30. Anthophyllite replaces biotite at Hiltuspuro. Field of view c. 2 mm wide. Plane polarized light. Photo by M. Pajunen.

cording to Luukkonen (1992), the crust had largely been cratonized before the Neoarchaean deformation events, D_4 , D_5 and D_6 . Cooling had advanced to a low temperature and crust rigidity was high. Low-temperature epidote-filled fractures are crosscut by Palaeoproterozoic diabase dykes in several places. The kyanite-bearing mineral assemblages penetrate the stabilized Archaean structures. Similar structures with metasomatic alteration also cut the amphibolite-facies metamorphic fabrics in the Palaeoproterozoic metadiabase dykes. Thus, the generation of kyanite rocks represents a Palaeoproterozoic reactivation event in the Archaean crust of eastern Finland.

Early-stage P-T conditions

The peak P-T conditions of the early stage are constrained by stabilization of the kyanite-staurolite assemblages. The increase in temperature is confirmed by reactions from Ms-Chl-Qtz to St-Ky. The minimum pressure of kyanite-staurolite stabilization is about 4.5–5 kbar on the basis of the andalusite-kyanite stabilities of Holland and Powell (1990) and the staurolite-in reaction of Winkler (1979). The mineral assemblages at Havukkalammit thought to represent early-stage conditions were tested with TWEEQU software (Berman 1991) using the thermodynamic dataset of Holland and Powell (1990). The assemblages did not, however, show equilibrium conditions, probably due to the late-stage reorganization of components.

The isochore for the primary fluid inclusions in kyanite is located in the stability field of andalusite-sillimanite (Fig. 31). This may indicate that fluid pressure was lower than lithostatic pressure (Walther 1990) during the entrapment or later reequilibration of the inclusions. At Levävaara, andalusite and kyanite crystallized simultaneously, implying that kyanite crystallized under conditions close to those of the kyanite-andalusite reaction boundary. In contrast, at Turkkivaara and east Havukkalammit, andalusite postdates kyanite. The dissimilar crystallization sequences of aluminium silicates in various targets reflect slight differences in crystallization pressures between targets. It is possible that at Turkkivaara and east Havukkalammit, andalusite was not stabilized until the late stage, in equilibrium with cordierite. The upper temperature limit during the early stage can be established from the breakdown reactions, St+Qtz and Chl+Qtz, and the polymorphic change of kyanite to sillimanite at about 600°C (Fig. 31).

Late-stage P-T conditions

The reaction history of kyanite rocks demonstrates that metasomatism occurred during the progression from greenschist to amphibolite facies conditions. The increase in temperature is well documented by prograde reactions from kyanite and andalusite stability fields to the sillimanite stability field (Figs. 8 and 27).

Decomposition of staurolite and quartz to cordierite and sillimanite is reliable evidence of an increase in temperature. Reaction (16) was experimentally studied by Richardson (1968) for the Fe-Al-Si-O-H system. In a Mg-free system the reaction proceeds at low pressures (2-2.5kbar/ 600°C), but Richardson (op. cit.) expects the stability field of Mg-bearing staurolite with quartz to be wider in temperature and, especially, to shift towards higher pressure. Here, the reaction was studied as an isochemical one (cf. p. 103) with TWEEQU software (Berman 1991) using the thermodynamic dataset of Holland and Powell (1990). The result (reaction A in Fig. 31) shows that the pressure estimate at 600°C coincides well with the other data. X_{H2O} is estimated to be 1.0 in all TWEEQU calculations. It is probable that the actual value was somewhat lower, but its effect on geological interpretation is small. The uncertainties in estimations are discussed in Holland and Powell (1990) and Berman (1991).

The chlorite breakdown reaction with quartz (26) produced anthophyllite-gedrite fibroblasts (Fig. 29) in which gedrite and anthophyllite occur as separate grains. The stability field of the Chl-Qtz assemblage was studied by Fleming and Fawcett (1976). They found that, for a Mg-rich composition, the pair is stable up to the kyanite-sillimanite reaction boundary at a pressure of about 4 kbar and is thus stable at the temperature achieved during the early stage. A crest in the



Fig. 31. Palaeoproterozoic P-T evolution of kyanite rocks at Hiltuspuro and Tetrilampi. The path is based on temperature determinations (Grt-Bt/Chl; Table 3), on TWEEQU analysis of assemblage Crd-Ath-Chl-Qtz-Ky/Sil and reactions A-D (see text), and on fluid inclusion data (P = primary inclusion, PS = pseudosecondary inclusion and S = secondary inclusion; Table 1). Aluminium silicate stabilities are from Holland and Powell (1990) and staurolite stability from Winkler (1979).

miscibility gap between the anthophyllite and gedrite series (Robinson et al. 1982) is estimated to lie at about 600°C in samples with a bulk $X_{M_{e}}$ of 0.6 (Spear 1980). The pressure dependence of the solvus has not been established, but the critical curve is suggested to have a positive dP/dT (Crowley & Spear 1981). On the other hand, the production of two orthoamphibole generations in one reaction and the number of reactions with only one phase imply that the reaction took place close to the temperature of the solid solution gap. The outcome of the TWEEQU calculation of the Mgrich assemblage, Ath-Crd-Am-Ky/Sil-Qtz, is presented in Fig. 31 with estimated X_{H2O} values 1.0 and 0.8. The variation in values, c. 610-650°C/ 4.5-5.0 kbar, is small and mainly in temperature. At the temperature given by the garnet-biotite/ chlorite thermometer, the calculated value of X_{H2O} is approximately 0.7-0.8. The effect of the variation in X_{H2O} on pressure estimation is insignificant for geological interpretations.

The garnet-quartz rock at northwest Hiltuspuro enables us to calculate temperature by thermometry based on Mg-Fe exchange in the mineral pairs chlorite-garnet and biotite-garnet. The analyses were made on a garnet partly replaced by chlorite and on a portion of a younger garnet generation rimming the chloritized part of the grain (Fig. 32). We attribute the growth of the garnet rims to the increase in temperature during latestage alteration - late-stage parageneses are well developed in adjacent outcrops. The most reliable estimates were obtained with the consistent program GEOPATH (Gerya & Perchuk 1992) (Table 3). The garnet core gives only an indicative average temperature of $638^{\circ}C$ (sx = 14.6°C), maximum 652°C. The rim garnet compositions give an average T of 602° C (sx = 11.6°C), maximum



Fig. 32. Garnet-chlorite-biotite assemblage in garnet-quartz rock at Hiltuspuro. Garnet (Grt_1) is replaced by chlorite, which is rimmed by later garnet (Grt_2) . Field of view c. 4 mm wide. Plane polarized light. Photo by M. Pajunen.

Table 3. Garnet-biotite and garnet-chlorite temperature estimates (G&P; Gerya & Perchuk 1992) of garnet-quartz rock at Hiltuspuro.

Grt-Bt	Analysis no.(*)	19&78	20&78	22&78	20&79	21&79	22&79
	Analysis point(**)	R-R	C-R	R-R	C-R	R-R	R-R
Grt	X _{Alm}	0.615	0.595	0.604	0.595	0.610	0.604
Grt	X_{Prp}	0.073	0.086	0.074	0.086	0.072	0.074
Grt	X _{Sps}	0.226	0.211	0.225	0.211	0.218	0.225
Grt	X _{Grs}	0.087	0.108	0.098	0.108	0.100	0.098
Bi	X_{Mg}	0.317	0.317	0.317	0.355	0.355	0.355
T°C	G&P	610	651	615	616	579	583
Grt-Chl	Analysis no.(*)	20&91	22&91	20&92	22&92	20&93	22&93
	Analysis point(**)	C-R	R-R	C-R	R-R	C-R	R-R
Grt	X _{Alm}	0.595	0.604	0.595	0.604	0.595	0.604
Grt	X _{Prp}	0.086	0.074	0.086	0.074	0.086	0.074
Grt	X _{Sps}	0.211	0.225	0.211	0.225	0.211	0.225
Grt	X _{Grs}	0.108	0.098	0.108	0.098	0.108	0.098
Chl	X _{Mg}	0.355	0.355	0.338	0.338	0.354	0.354
T°C	G&P	637	603	652	617	638	603

(*) analysis numbers in Appendices 5, 8 and 9.

(**) R = rim and C = core

 617° C. Pressure was estimated to be 5 kbar and $X_{\rm H2O}$ 1.0 in both calculations. The P-dependence of temperature is restricted and the estimated pressure coincides well with other observations, e.g. the kyanite-sillimanite transformation. The rim of a large garnet generation portion coexists with Ms-Pl-Qtz. The stability of the reaction

(32) $Grs^{Grt}+Pg^{Ms}+Qtz \rightarrow (Ab+An)^{Pl}+H_2O$

was calculated with TWEEQU (Berman 1990, 1991), using the thermodynamic dataset of Holland and Powell (1990). Reaction C (Fig. 31) runs very near the kyanite-sillimanite reaction boundary.

The late-stage structures are due to the pronounced effect of fluid – static replacement reactions along microfractures in grain boundaries, for instance. The pseudosecondary fluid inclusions in cordierite record c. 4 kbar pressure at the peak temperature; this is about 1 kbar lower than suggested by the kyanite-sillimanite transformation and reactions A, C and D (Fig. 31).

At the upper crustal levels of brittle behaviour, fluid pressure is hydrostatic, $P_f = P_h$, but at greater depth it approaches lithostatic pressure, P1, that is suprahydrostatic fluid pressure or fluid overpressure, because of decreasing porosity (e.g. Gavrilenko & Gueguen 1993). At great depths, beneath the level of brittle-ductile transition of the crust, fluid pressure is proposed to be lithostatic, $P_f = P_1$ (Walther 1990). Thus, a layered division of crustal fluid convection and P_f regime is suggested (e.g. Etheridge et al. 1983, Oliver et al. 1990, Ivanov & Ivanov 1993). Supralithostatic fluid pressures, $P_f > P_l$, may be maintained only briefly and locally (see Sibson 1992, Gavrilenko & Gueguen 1993), because of the tensile fracturing that may occur when P_f exceeds the minimum principal compressional stress, σ_3 , of the rock (e.g. Etheridge et al. 1983).

It is unlikely that the medium-pressure assemblages in the kyanite rocks were formed due to high fluid pressure caused by fluid entrapment, because there are no signs of such a widespread trap system in the Nurmes–Sotkamo area. Moreover, the high density contrast between the fluid

and tonalitic country rock forces the hot fluid to upwell at P_f lower than P₁ (cf. Lister & Kerr 1991). The high temperature supercritical fluids have very low viscosities and therefore also high mobilities (Wickham 1992). Some structures, such as microfracturing and subsequent metasomatism along grain boundaries, suggest fluid agility in places, but high fluid overpressures causing a systematic metasomatic-metamorphic reaction series such as that in kyanite rocks is not plausible. The early-stage structures are well preserved from the later violent brecciation that could be expected to form if high fluid pressures should relax. Thus, as we see it, the pressure obtained, 4-5 kbar, is really lithostatic, and it is even possible that fluid pressure followed its own path at lower pressure. This interpretation is supported by metamorphic assemblages suggesting corresponding crystallization pressures in the neighbouring Palaeoproterozoic North Karelia Schist Belt (Koistinen 1981, Treloar et al. 1981, Halden & Bowes 1984) and Kainuu Schist Belt (Tuisku & Laajoki 1990, Tuisku 1991). Because all the reactions described (e.g. A, C and D in Fig. 31) are to some extent dependent on pressure and because the lower pressure assemblages have a clear tendency to stabilization, some exhumation must have occurred between the early and late stages. The pressure estimate implies that, during the late stage, the Nurmes-Sotkamo area was thickly covered by Archaean ± Palaeoproterozoic crust. The calculated depth ($P = \rho * h * g$, where $\rho = \text{density}$, h = height and g = gravitationalacceleration) is about 15.7-19.6 km on the basis of the average density of rocks, 2600 kg/m³, in the Nurmes-Sotkamo area (see Yliniemi et al. 1993), or about 11.7 km on the basis of fluid inclusion isochores and assuming that the pressure of inclusions represents lithostatic pressure.

The mineral assemblages of metasomatic rocks imply an overall increase in metamorphic grade southwards. Thus, the late thermal event, characterized by CO₂-rich fluid, was more pronounced in the south, but was not absent from the north either, e.g. Tuomaanvaara, Turkkivaara and east Havukkalammit. According to the microstructure it is possible that andalusite stabilized with cordierite at east Havukkalammit and Turkkivaara during the late stage. At Turkkivaara, the temperature continued to rise in the sillimanite field, but at east Havukkalammit the thermal peak was achieved under andalusite field conditions, suggesting a slight difference in peak temperatures between occurrences.

ISOTOPE GEOLOGY

Radioactive mineral inclusions causing a yellow halo in cordierite are typical of the cordierite-bearing samples (Fig. 33). A coarse-grained kyanitechlorite-cordierite rock sample (A1298) from southeast Hiltuspuro (see Figs. 2b, 3 and 5) was separated for isotopic study. Xenotime, which becomes enriched in the heavy, >4.3 g/cm³, 1.4 A, magnetic Franz fraction, occurred as fresh, yellowish or greenish grains with few crystal faces. The length/width ratio of the crystals, 70–160 μ m in diameter, was close to 1. The majority of the grains were translucent, but cloudy, poikiloblastic grains also occurred. Two U-Pb analyses (Table 4) were made in the Isotopic Laboratory of the Geological Survey of Finland in Espoo. Concordant analyses gave a very accurate age of 1852 ± 2 Ma (Fig. 34). A very high U content and low common lead content are characteristic of the analyses.

The crystallization of the cordierite-sillimanite paragenesis represents metamorphic peak conditions of kyanite rocks at about 600-620°C/ 4-5 kbar. Xenotime is intimately associated with these highest grade phases, indicating that the radioactive elements, and REE and the closely related yttrium, were mobilized under approximately such conditions. Enrichment of the radioactive components, U and Th, in shear zones has been observed by some workers, e.g. Kerrich et al. (1991). According to Aleinikoff and Grauch (1990), REE transport in fluid occurs in carbonate, phosphate or sulphate complexes or in fluorine-rich fluids. During late-stage alteration, the fluid was rich in CO₂, and phosphates, apatite and xenotime are fairly common minerals. Transport of REE as carbonate and phosphate complexes is



Fig. 33. Xenotime and apatite in cordierite-rich assemblage at Hiltuspuro. Field of view c. 2 mm wide. Plane polarized light. Photo by M. Pajunen.

Table 4. U-Pb data on xenotime of kyanite-chlorite-cordierite rock (sample A1298) at Hiltuspuro (x=7070.51 and y=4485.25).

Sample				Meas	sured	1)	1)	2)	2)	Age	(Ma)
	density / size	²³⁸ U	²⁰⁶ Pb	²⁰⁶ Pb/	²⁰⁷ Pb/	²⁰⁸ Pb/	²⁰⁶ Pb/	²⁰⁷ Pb/	²⁰⁶ Pb/	²⁰⁷ Pb/	²⁰⁷ Pb/
	(g/cm ³) (µm)	(ppm)	(ppm)	²⁰⁴ Pb	²⁰⁶ Pb	²⁰⁶ Pb	²³⁸ U	²³⁵ U	²³⁸ U	²³⁵ U	²⁰⁶ Pb
A1298A	>4.3 / >70	9483	2653	90608	.1134	.1046	.32331	5.0504	1805	1827	1853
A1298B	>4.3 / >70	10748	3069	75066	.1133		.33000	5.1519	1838	1844	1852

¹⁾ Corrected for blank (0.5 ng Pb).

²⁾ Corrected for blank and common lead (Stacey & Kramers 1975).



Fig. 34. Concordia diagram of xenotime of coarse-grained kyanitechlorite-cordierite rock from the outcrop shown in Fig. 3 at Hiltuspuro (x=7070.51, y=4485.25). Sample site (A1298A–B) also indicated in Figs. 2b and 5.

thus possible. For the Hiltuspuro carbonate minerals, this would imply precipitation of carbonate complexes; these are, however, rare. Instead, the abundant phosphates suggest that phosphate complexes were the most probable media of transport and deposition. Even slight changes in pH result in phosphate destabilization and deposition (Mineyev 1963).

Schärer et al. (1990) reported xenotime and monazite ages on a leucogranitic intrusion in the Ailao Shan-Red River shear zone in China. They found no significant differences between the dates and concluded that the U-Pb closure in xenotime corresponds to that of monazite. Monazite closure temperature estimates vary between 530°C (Wagner et al. 1977) and 720–750°C (Copeland et al. 1988). Observations of Smith and Barreiro (1990) suggest that monazite is immune to lead loss and resetting of the U-Pb clock at or below upper sillimanite grade metamorphism. Thus, the closure temperature of xenotime is thought to be close to the temperature obtained from Hiltuspuro. Smith and Barreiro (1990) found that metamorphic monazite (xenotime?) may even crystallize under staurolite-grade conditions, at above $525\pm25^{\circ}$ C and 3.1 ± 0.5 kbar. The faint signs of retrograde processes after the peak conditions, at Hiltuspuro in particular, mean that the fluid activity and crustal cooling, and, thus, the U-Pb blocking of xenotime, terminated rather rapidly. Therefore, the age 1852 ± 2 Ma refers reasonably well to the time of the thermal peak.

DISCUSSION

Confirming the actual metasomatic nature of a rock without studying the relation of the metasomatic process to tectonic structures is often difficult; in other words, we have to verify that the alteration overprints an earlier tectonic-metamorphic feature. It is important to distinguish between metasomatism and syngenetic alteration, especially, when interpreting genetic features of ores. The diverse theories formulated on the genesis of cordierite-anthophyllite rocks and the base metal sulphide ores related to them or in their vicinity are a good example.

Structural and metamorphic studies show that the Archaean crust was largely cratonized and that cooling was far advanced during the last Neoarchaean tectonic events. As illustrated by Figs. 4, 9, 12 and 16, the kyanite rocks crosscut the regional metamorphic structures in the Archaean tonalitic rocks. Thus, their formation represents a new tectonic, metamorphic and metasomatic episode and reactivation of the Archaean crust.

The assemblages in kyanite rocks show metasomatism under prograde metamorphic conditions, from Chl-Ms-Qtz to Ky-St and further to Crd-Sil stabilities. The earliest assemblages of kyanite rocks represent lower-grade metamorphic conditions than do those in their host rocks. The differences in metasomatic mineral growth successions in various locations indicate fluctuations in metasomatic-metamorphic conditions. The differences are clearest in Al-silicate crystallization sequences: Ky (Kallioniemi, Kauhealampi and Korpilampi), Ky-Ky (Havukkalammit), Ky-And (east Havukkalammit), Ky-And-Sil (Turkkivaara), Ky/

And-Sil (Levävaara) and Ky-Sil (Hiltuspuro, Teljo and Tetrilampi) (Fig. 35). It is obvious that in shear, fault and fracture zones metasomatic-metamorphic reactions proceed to completion in some parts of the zone whereas in others the rock may be totally unaffected. The progression of reactions also depends on the structural state of the reactive zone, i.e. on the openness of the zone for fluids. Besides, it is unlikely that analogous reactions always took place simultaneously in different zones. Even if no mineral reactions can be detected, phases may be restabilized by exchange reactions. The P-T paths of the zones studied cannot be directly correlated with each other, and yet their similar clockwise forms suggest that the metamorphic succession of the zones and crustal evolution of the areas were similar.

Fluid composition

The fluid inclusion data on Hiltuspuro are informative about the fluid composition during metasomatism. The primary fluid inclusions in early kyanite are dense and saline (11-17% weight equivalent NaCl) H₂O (Table 1), and represent a fluid that was trapped during kyanite growth. Very high salinity fluids (up to 40% weight equivalent of NaCl from minerals) have been described from deep-penetrating shear zones in Alpine areas (personal communication of S. Tempest reported in McCaigh 1989). More normal values are about 20-25 wt% dissolved salts. The salinity of groundwater, which increases with depth in the Canadian Shield, is postulated to be caused by rock-fluid interaction at greater depth (Frape & Fritz 1987, Frape et al. 1984). Isotopic studies suggest that meteoric water may even reach depths as great as 20 km (see Nur & Walder 1990). The early-stage fluid in kyanite rocks may have been expelled from an overthrust Palaeoproterozoic sequence soon after its collision. This meteoric water percolated deeper into the crust. The high salinity of these fluids is hard to explain without fluid-rock interaction.

During the late stage, the existence of a greater proportion of CO_2 in fluid phase is demonstrated by the pseudosecondary fluid inclusions in



cordierite. As no carbonates were formed in this high temperature process, opportunities for estimating fluid composition by physico-chemical methods are limited. H₂O-rich fluids in the deep crust, at above 650°C, dissolve in silicate melts (e.g. Fyfe 1973, Newton 1990). An increase in migmatization with a concomitant increase in CO₂ in fluid inclusions has been shown to have occurred in the low-pressure Svecofennian metamorphic belt in southern Finland by Poutiainen (1990). CO₂ infiltration is a major factor in granulitization of the lower crust (e.g. Newton et al. 1980, Harris 1989). These fluids may derive from mantle degassing processes or magmatic production of CO₂ (Vry et al. 1988). Observations of CO₂-bearing fluid causing late metamorphism have been reported by Gruau et al. (1992) from the southern part of the Archaean Kuhmo Greenstone Belt. Cordierite-sillimanite metamorphism is pronounced in the southern targets, but cordierite patches very similar to those at Hiltuspuro also exist in the northernmost target, Tuomaanvaara. We assume that cordieritization - Mg-metasomatism and CO2-rich fluidization - was, indeed, very far-reaching and also suspect that the CO2-rich

fluid of the late stage had a deep-seated origin – e.g. mantle degassing or magmatic CO_2 – over a wide area.

The Palaeoproterozoic fluid compositions at Hiltuspuro differ markedly from the fluid inclusions in quartz veins in the Archaean Moukkorinaho Au mineralization in the Suomussalmi Greenstone Belt (north of the area shown in Fig. 1); there the mineralizing fluids were low-saline complex H_2O - CO_2 mixtures (Luukkonen et al. 1992, Poutiainen & Luukkonen 1994). The ore mineral contents of the kyanite rocks in the Nurmes– Sotkamo area are very low and the metasomatic processes described may well have depleted, not enriched, the system in ore constituents.

Tectonic-metamorphic correlation

Early stage

The early-stage kyanite-staurolite metasomatismmetamorphism has not been dated. According to structural observations on the western occurrences, the kyanite-staurolite assemblage was formed by the immediate static state rise in temperature af-

North Kare	lia Schist Belt					
Koistinen (198	1), Outokumpu are	a	Campbell	et al. (1979), Heinäv	vaara area	Major
Deformation and trend	Mineral growth	P-T and timing	Deformation	Mineral growth	P-T	tectonics
D1	early - pre-D1 Bt	greenschist facies	D1	S1: mica schistosity		
NNE-NE	S1: Bt schistosity	Bt grade		stubby mica		
tectonic transport	post-D1 Grt					thrust
D2	S2: dominant Bt-Ms	temperature	D2	S2: mica schistosity	temperature	tectonics
NNE-NE	schistosity/crenulation	increases		Bt and Ms, strain-free	increases	
tectonic transport	post-D2 Bt growth					
D3 (D2c)*	S3: spaced Bt		not			transition
NW-SE,	schistosity/crenulation	middle amphibolite	present	post-D2 St	metamorphic	
rare	post-D3 St	facies about			peak :	
NNE-SSW	Ky juxtaposed with St	1860 Ma ago		pre-D4 Grt	$T = 675 \pm 25$ °C	~
D4 (D3)	S4: crenulation	Treolar et al. 1981:	D4 (D3)	S4: zonal crenulation	$P = 5 \pm 0.25$ kbar	srike slip
145-185°	St rim on S4	$T = 600 \pm 50^{\circ}C$		syn-D4 Grt	PH2O = 0.6 Ptot	and
	some new Bt	$P = 3.5 \pm 1 \text{ kbar}$		syn/post-D4 Ms and Bt		transform
	St inclusions in And -			Sil on F4 fold hinges		fault
	stage ?					tectonics
D5 (D4)	S5: weak cleavage		D5 (D4)	S5: fracture cleavage		
225-245°				Ms and Chl growth		
	late Chl - stage ?	retrograde			retrograde	
D6 (D5)	S6: fracture or	metamorphism	D6 (D5)	S6: fracture cleavage	metamorphism	
080-100°	healed cleavage			Chl and Ms growth		
D7(D6)	S7: fracture cleavage					
000-015°	Chl, pseudotachylites					

Table 5. Correlation scheme of tectonic-metamorphic successions between the Palaeoproterozoic North Karelia

* Deformation names used by the referred authors are in parantheses

ter the early shearing that generated the foliated chlorite-muscovite-quartz assemblages. The present data do not exclude the possibility that, in some places, the earliest alteration is a product of a Neoarchaean or pre-collisional Palaeoproterozoic (e.g. related to diabase evolvement) event. However, the successive series of metasomatic-metamorphic reactions in kyanite rocks and the structural relationships between kyanite rocks and metadiabase dykes, e.g. at Levävaara, confirm that they belong to the same Barrovian-type progression.

Correlation of the metamorphic succession of kyanite rocks to that of the Palaeoproterozoic metasedimentary sequences in the Kainuu and North Karelia Schist Belts gives an idea of its date (cf. Campbell et al. 1979, Koistinen 1981, Tuisku & Laajoki 1990, Tuisku 1991); the main structural and metamorphic stages in the schist belts are summarized in Table 5. The first deformation phases, D_1 and D_2 , were formed by extended thrusting of the allochthonous Palaeoproterozoic slices over the Archaean craton and autochthonous sequences (see Koistinen 1981, Tuisku 1991). The dominant mica schistosity in metasediments was formed. D_3 represents the transitional stage from thrust to transform fault tectonics. Regional metamorphic peak conditions were achieved in both areas after D_3 , at about 1870–1860 Ma ago. At that time deformation was characterized by strike-slip

Kainuu	Schist Belt			Archaea	n Nurmes-Sotka	mo area
Tuisku (19	91), Puolankajärvi area		Major	This study		
Deformation and trend	Mineral growth	P-T and timing	tectonics	Deformation and trend	Mineral growth	P-T and timing
D1 NE tectonic	S1: segregation layering syn-D1 Grt	T = c. 400 °C		Collision (not identified	Park 1988 and	Vaasjoki and Sakko 1988:
transport	paragenesis: Grt-Chl-Bt	temperature	thrust	in this study)	Park and Bowes 1983:	about 1.9-1.875 Ga
D2		increases	tectonics		shearing in major	ago
E tectonic					D2 zones	
transport				DP1	metamorphism and	timing unknown
D3	S3: crenulation or segregation	$T = 535^{\circ}C$	transition		deformation of diabases	(pre- or syn-collisional ?)
	segregation cleavage	and following		Early stage	ductile shearing	
	paragenesis: Grt-Bt-St. last	isothermic		NW-SE	mobile brecciation	temperature
	regional metam. assemblage	decompression			Ms-Chl-Qtz	increases
D4	S4: mica schistosity	metamorphic peak:	srike slip		static Ky-St	$T = 550-600^{\circ}C$
NE-SW	paragenesis: Grt-Als-Bt-St	$T = 550-580^{\circ}C$	and		±And, Cm	P = about or
	Ky earlier than Sil	P = 4-6 kbar	transform			in excess of 5 kbar
		c. 1870 Ma ago	fault	Late stage	semi-brittle fracturing	peak metamorphism
	Crd+And during or after D4	decompresssional	tectonics	DP2	static growth	T = 600-620 °C
D5	S5: crenulation cleavage	cooling to		NW-SE	Crd-Sil±And	P = 4-5 kbar
N-S	syn-D4 or later Crd-And	$T = 500-530^{\circ}C$		rare N-S	Oam	1852±2 Ma ago
		P = 2.5 - 3.5 kbar		or NE-SW		
D6	S6: fracture cleavage			Cooling	late fractures	rapid cooling
E-W				stage	Chl, Ms, Crb	

and Kainuu Schist Belts and kyanite rocks in the Archaean Nurmes-Sotkamo area.

or transform fault tectonics. The shear zones that formed during these later stages, e.g. the Auho shear zone, crosscut the Archaean rocks (Tuisku 1991). Staurolite crystallized during static heating after D_3 , and in North Karelia the staurolite rim grew on S_4 schistosity (Koistinen 1981). Kyanite and sillimanite were stabilized: in the Kainuu Schist Belt kyanite preceded sillimanite (Tuisku 1991), and in the North Karelia Schist Belt kyanite crystallization was juxtaposed to staurolite growth (Koistinen 1981). Metamorphism was due to isothermal decompression between D_3 and D_4 . Both schist belts show andalusite and/or cordierite growth after staurolite and, according to Tuisku (1991), their stabilization represents decompres-

sional cooling after D_4 . Later deformation phases record retrograde conditions, as evidenced by chlorite and muscovite overgrowth on the peak metamorphic assemblages in late shear or fracture zones.

Although structural correlation between the Palaeoproterozoic and Archaean Nurmes–Sotkamo areas is not straightforward, their metamorphic P-T successions were very similar (Table 5). The early shearing observed in kyanite rocks may be correlated to $D_{3/4}$ shear zones in the Palaeoproterozoic areas, but we have insufficient data to confirm this. Our interpretation is that the early-stage kyanite-staurolite metamorphism in the Archaean area largely corresponds to the Palaeoproterozoic, post-D₃ or D₄, kyanite-staurolite metamorphism in the Kainuu and North Karelia Schist Belts and, thus, was formed during the post-tectonic stage with respect to the Palaeoproterozoic collision and crustal thickening. Similar metamorphism in the eastern Archaean area in Russia is also interpreted as Proterozoic in age (e.g. Belyaev et al. 1993, Petrov 1993). The pressure obtained here, ≥ 5 kbar, is lithostatic, meaning that this crustal level was overlain by a thick pile of Archaean and/or Palaeoproterozoic rocks.

The Palaeoproterozoic metadiabase dykes are often undeformed and carry only static metamorphic assemblages recording amphibolite and epidote-amphibolite facies conditions. Some more strained dykes show well-developed metamorphic foliation. At Hiltuspuro and Levävaara, the amphibolite-facies metadiabase dykes are foliated, and have a steep southwesterly to westerly dip. The foliation of metadiabases can sometimes be traced into the tonalitic host rock, but correlation further away from the contact zone is uncertain. The signatures of shearing and schistosity development that are visible in some dykes but not in others demonstrate that Palaeoproterozoic deformation, although still operational rather far east in the Archaean area, was there restricted in zones. The timing of the amphibolite-facies foliation in metadiabase dykes has, however, still to be established. Clearly the dykes were affected by two separate epidote-amphibolite to amphibolite facies events: the earlier penetrative and the later zonal with metasomatism. Whether the earlier event is connected to the early extension of the Archaean "supercontinent" (Lahtinen 1994) or to the Svecofennian collision is not known.

Late stage

In the Palaeoproterozoic Kainuu Schist Belt the cordierite- and andalusite-bearing assemblages were stabilized during the post- D_4 and D_5 stages (Table 5). There are very similar metamorphic assemblages in some western kyanite rocks. According to Tuisku (1991), they indicate decompressional cooling, whereas our observations of kyanite rocks show that the late cordierite-silli-

manite/andalusite paragenesis represents the thermal peak during decompressional heating, suggesting an extensional stage (cf. Platt & England 1993).

Before our study, no age determinations had been conducted on an approximately 1850-Ma-old process that is, furthermore, exactly linked to tectonic-metamorphic structures in the Archaean area. On the other hand, comprehensive age data on the Palaeoproterozoic area show that only a small proportion of post-tectonic magmatism happened between 1880 Ma and 1850 Ma (Huhma 1986, Vaasjoki & Sakko 1988). The nearest rocks dated to that age group are in the North Karelia Schist Belt - the post-tectonic Maarianvaara granodiorite (Fig. 1), which is 1857±8 Ma old (Huhma 1986) and in the Kainuu Schist Belt - the Ristijärvi granodiorite, which is 1859±8 Ma old (Kontinen & Meriläinen 1993). Granitic intrusions of this kind are rather common to the west of the study area, in a north-south-trending zone near the Archaean-Palaeoproterozoic boundary. According to the Sm-Nd isotopic composition, these magmas represent a mixture of Archaean and Palaeoproterozoic material (Huhma 1986). Granodiorite of similar age, 1.853±8 Ga (Vaasjoki & Sakko 1988), has been described also further west in Svecofennian terrain. The existence of these granitoids is an evidence of marked thermal input into the lower crust at about 1.850 Ga ago.

Huhma (1981) described microtonalitic dykes crosscutting the Maarianvaara granodiorite. Microtonalites are widely distributed in the North Karelia Schist Belt. He interpreted their sphene U-Pb age, 1850 Ma, as metamorphic, but crosscutting relations place their age between 1857±8 Ma (Maarianvaara) and 1830 Ma, which is the age of local lamprophyre dykes (Huhma 1981). It is interesting that the lamprophyre dykes are characterized by high apatite and LREE contents (Laukkanen 1987) - geochemical features that seem to be characteristic of late-stage kyanite rocks, too. These widespread thermal and magmatic events are of the same age as the metamorphic peak in kyanite rocks, suggesting the same energy source for the processes.

Resetting of the K-Ar isotopic compositions of

biotite and hornblende in the Archaean TTG-granitoids indicates an increase in temperature during the Palaeoproterozoic era (Kontinen et al. 1992, O'Brien et al. 1993). Palaeoproterozoic reheating is also postulated by Gruau et al. (1992) and Karhu et al. (1993).

Post-tectonic magmatism in Karelian terrain shows that the Palaeoproterozoic+Archaean crust underwent some kind of disturbance between 1860 and 1830 Ma ago. It is possible that reworking and melting in the lower crust and upper lithosphere (cf. Fountain 1989) were then more pronounced than can be inferred from the current erosion level. The event may be connected with post-collisional extension of the upper crust (cf. Gaudemer et al. 1988), as suggested by dilatational structures and the decrease in pressure in kyanite rocks and also by the decompressional heating history. The process produced enough CO2-rich fluid and energy for the heating and metasomatism detected. Extension and upwelling of the crust by this deepcrustal process caused rapid tectonic and/or erosional exhumation, termination of fluid action and crustal cooling. The crust, especially the high-velocity layer of the lower crust, of eastern Finland is exceptionally thick (Yliniemi et al. 1993, Korja et al. 1993). Korja et al. (op. cit.) attribute this to Palaeoproterozoic reworking of the lower crust and the Moho. The results of our study support this concept. The thermal event and the characteristics of the metasomatism are interpreted to be connected with magmatic underplating that transformed the lower crust up to the Kuhmo Greenstone Belt in the east during Palaeoproterozoic time.

Because the Palaeoproterozoic structures identified in the eastern part of the Archaean area are zonal and rather limited, the old stabilized crust must have behaved very competently and in large coherent slices. The impact of metamorphism on these slices and blocks was slight and static in type, because the TTG crust was cool and dry. The energy that would have been needed for largescale penetrative metamorphism was apparently not available. The most intense Palaeoproterozoic metasomatic and metamorphic overprints are therefore visible along the zones of weakness, which were open for an intense fluid flow capable of transporting sufficient heat to produce the features seen in kyanite rocks.

CONCLUSIONS

The main conclusions drawn from this study are:

- 1. Palaeoproterozoic deformation zones crosscutting the Archaean crust in eastern Finland show a change in deformation style from early-stage ductile to late-stage semi-brittle, and have been affected by strong metasomatism.
- The early-stage metasomatic-metamorphic event shows progression from Chl-Ms-Qtz to Ky/And-St stability of Barrovian-type assemblages, presumably related to crustal thickening during the Svecofennian orogeny and to the temperature increase soon thereafter. Metasomatism was induced by a saline H₂O fluid.
- The late-stage semi-brittle event exhibits dilatational features. Mg metasomatism at increasing temperature and CO₂-rich fluidization produced cordierite-rich assemblages in the sillimanite or andalusite stability field under maximum P-T conditions of about 600–620°C/4–5 kbar. This peak metamorphism is dated to 1852±2 Ma.
- 4. The metamorphic assemblages in the metasomatic zones indicate small pressure differences in the Nurmes–Sotkamo area, i.e. in the thickness of the column of overlying Palaeoproterozoic±Archaean rocks, about 1850 Ma ago. The thickness of the exhumed slice was about 15–20 km. The pressure decrease indicates slight exhumation between the early and late stages, and the late-stage heating was decompressional. The dilatational structures and decompressional thermal evolution indicate extensional evolution during the late stage.
- 5. The thermal event at 1850 Ma is a distal reflection of high overall thermal activity in the Palaeoproterozoic Svecofennian area in the west but, here, in the Archaean craton, it was restricted to fracture zones and was connected to the evolution of the lower crust with a thick high-velocity layer. Thermal input is interpreted to be a result of magmatic underplating.

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Appendix 1. Mineral abbreviations used in text.

Ab	= albite
Alm	= almandine
Aln	= allanite
Am	= amesite
An	= anorthite
And	= andalusite
Ap	= apatite
Ath	= anthophyllite
Bt	= biotite
Cam	= clinoamphibole
Cel	= celadonite
Chl	= chlorite
Crb	= carbonate
Crd	= cordierite
Crn	= corundum
Ep	= epidote
Ged	= gedrite
Grs	= grossularite
Grt	= garnet
Hbl	= hornblende
Ilm	= ilmenite

Ky = kyanite **Mag** = magnetite Ms = muscovite **Oam** = orthoamphibole **Olg** = oligoclase Pg = paragonite Pl = plagioclase **Prp** = pyrope Py = pyrite **Pyh** = pyrrhotite Otz = quartzRt = rutile Sil = sillimanite Sph = sulphide**Spn** = sphene Sps = spessartine St = staurolite **Tou** = tourmaline **Xen** = xenotime + unidentified radioactive mineral Zrn = zircon

Ksf = potassium feldspar

Appendix 2. Mineral assemblages of the rock types studied. Intensity of metasomatism is classified as none = unaltered, weak = weakly altered, but primary structure well preserved, strong = penetratively altered, but primary structure still identifiable and penetrative = old phases occur only as relics in totally recrystallized and neomineralized rock. X = major mineral, o = minor mineral and + = accessory mineral. Pz = Palaeoproterozoic.

Northwestern occurrences	Mineral abbrev	viatio	ons a	are	liste	d in	Ap	pen	dix 1	1.																	
Occurrence (in Fig.1)/rock type	Alteration	QZ	P	Ksf	B	Chi	Ms	Cam	Oam	Crd	St	Grt	Ку	Sil	And	Ep	Aln	Spn	Zrn	Xen*	Ap	Tou	Rt	Mag	IIm	Sph	Crb
Tuomaanvaara (1)																											
Tonalite	weak	X	X		X	0											+		+		+						
Chl-Qtz rock	penetrative	X			0	X																					
Bt rock	penetrative	+			X	+																					
Crd-Chl-Bt rock	penetrative	0	0		0	0	+			X							+		+		+						
Turkkivaara (2)																											
Bt-PI-And rock	penetrative		X		X	+	+							+	X				+								
And-Ky-Crd rock	penetrative	Γ	X		0	0				X			0	+	X				+						+		
Bt-Crd rock	penetrative				X		0			X									+	+							
Bt-Crd-St rock	penetrative				X	+				X	0		0						+	+	+		+		+		
Kallioniemi (3)		Г																									
Tonalite	none	X	X		X	+													+		+						
Ky-bearing tonalite	strong	X	X		X	0	0						+						+		+		+				
Chl-Ms-Ky rock	penetrative	X	0			X	X						0						+		+		+				
Varpuniemi (4)		Γ																									
St-Chl-Pl rock	penetrative	+	X		+	X	0				X		0						+		+		+				
Korpilampi (5)		Г										-															
Ky-Bt-PI rock	penetrative	0	X		X	0							0						+		+		+				
Kauhealampi (6)			3																								
Ky-Chl-Ms rock	penetrative	X	X			X	X						0						+		+		+				
Havukkalammit (7)		Г			1.0																						
Mag tonalite	strong	Г	X		X	0										+			+		+			X			
Ep rock	penetrative	Γ	0		0			0								Х							+	0			
Ky-PI rock, coarse	penetrative		X		X	0	+						X						+		+						+
East Havukkalammit (7)		Г		-	-		-			-			-														
Crd-Bt-PI rock	penetrative	0	X		X	0	+			0									+		+						
Crd-Ky-And rock	penetrative				X	0	0			X			X		X					+	+		+				
Matovaara (8)																											
Oam-mica rock	strong	0	X		X	+	+		X										+		+					+	

* and unidentified radiogenic minerals

Appendix 2. cont.

Southeastern occurrences	Mineral abbrev	iatio	ns a	are I	iste	d in	Ap	pen	dix 1	۱																		
Occurrence (in Fig.1)/rock type	Alteration	017	P	Ksf	₽	Chi	Ms	Cam	Oam	Crd	St	Grt	Ку	Sil	And	Cm	Ep	Aln	Spn	Zrn	Xen*	Ap	Tou	Rt	Mag	IIm	Sph	Crb
Levävaara (9)																				_								
Metadiabase, Pz	none	0	X						X													+				0		
Tonalite	weak	0	X		х	0	+													+	+	+		0				
Tonalite	strong	0	X		0	X	+													+	0			0				
Tonalite, mylonitic	weak	Х	Х		0	0	+													+				0				
Bt-Chl rock, foliated	penetretive		+			Х	х													+	+		0	0			0	
Tur-ChI-PI rock, coarse	penetrative		X			X														+	+		0	0				
Chl-Pl-Ms rock, coarse	penetrative	+	X			X	X													+	+			0				
Ky-Chl-Ms rock, coarse	penetrative	+	0			X	Х						X							+	+			0				
Chl-Pl rock, coarse	penetrative	0	х		+	X	+													+	0			0				
And-Ky-Chl rock, coarse	penetrative	0	X			X	X						0		X					+	+			0				
Crn-Bt-Chl rock, coarse	penetrative		0		X	X	0						0			0				+	+			0			+	
Ky-Crd-Chl rock, coarse	penetrative		X		х	X	0			0	0		X	0						+	0	0		0				
Ky-Chl-Pl rock, coarse	penetrative		X		+	X	0				0		0							+	0			0				
Hiltuspuro (10)												_			-	-	-											
D3 Hbl-tonalite	none	х	X		X			X									+	+	0			+						+
D3 Bt-tonalite	none	Х	X	0	X	0		0									0		+	+		+						+
Leucotonalite	weak	х	X		X	0														+		+		+				+
Tonalitic migmatite	weak	X	X		X	0														+		+		+		+		
Metadiabase, Pz	none	+	Х		+			X										+	0			+				0		
Metadiabase, Pz	none	0	Х		0			X										+	0			+						
Metadiabase, Pz, zone	strong	0	Х				+	X									0	+				0				0		
Ky-Crd rock, zone	strong	х	X		X	0	+			0			0	+						+				+				
Ky-Crd-St rock, zone	strong	х	X		X	+	+			X	0		0							+	0	+						
Crd-rock, spot	penetrative	х	0		0	0	+			X	0		0							0	+	+		+			0	
Crd-Oam rock, spot	penetrative	x	+		0	X	+		X	X			0							+	+			0		+		
Kya-Crd-St rock, spot	penetrative	0		-	X	+	+			X	X		X	+						+	0	0		+		+	0	
Crd-Oam rock, spot	penetrative	X	0	-	0	0	+		X	X			0									+		0				
Crd-Oam rock, spot	penetrative	x			+	+	+		0	X											+	+		0				_
Ky-Crd-Chl-Pl rock, coarse	penetrative		X		0	X	+			X			X	+						+	0	0		0				
Ky-Crd-Chl rock, coarse	penetrative	0	0			X	0	-		X			X							+	0	0		0				
Crd-Chl-Pl rock, coarse	penetrative		X		0	X	0			X										+	+			0				+
Bt-Chl rock, coarse	penetrative				X	X	+			X			0	+							+			+				
Teljo (11)			_																									
Tonalitic gneiss	none	х	X		Х	+	+										+			+		+						
D3 tonalite	weak	х	х		X	+														+	+	+		+				
Ky-Crd rock, zone	strong	0	X		X	0	+			X			X	0						+		+		0				
Ky-Crd rock, spot	strong	X	X		X	0	0			X			X	0						+	+			0				
Ky-Crd rock, spot	penetrative	х	X	-	0	+	+			X			X	+						+				0				
Egyptinkorpi (12)			_				-															-				-		_
Kva-PI rock	penetrative	х	X		+	+	+						X	0										+			+	
Tetrilammit (13)															-													-
Tonalite with Sil-zone	weak	х	X		X	+	0							X						+		+		+				
Ky-Crd rock, zone	strong	X	X		X	+	0			X			X			-				+	+							
Ky-Crd rock, zone	strong	X	X	-	0	+	0			X	_		0	0		-		_		+	0			0				
Ky-Crd-Sil rock, zone	strong	Х	X		0	+	0			X			X	X						+		+		0				

* and unidentified radiogenic minerals

Analysis no.	1	2	3	4	5	6	7	8	9	10	11	12	13	14
Occurrence	Hiltusp	uro:											Tetrilam	pi:
Sample/ring	1A1/1	1A1/2	1A1/2	1A1/3	1A2/1	1A2/2	1A2/3	10A/1	10A/2	10C	12A	12C1	15B2/1	15B2/2
n	5	1	1	2	1	1	2	1	1	1	2	2	3	2
SiO ₂	49.16	49.49	49.16	49.05	49.15	47.95	48.69	48.22	48.81	47.89	49.52	47.94	50.15	50.64
TiO ₂	0.01	0.00	0.01	0.01	0.00	0.01	0.02	0.00	0.01	0.01	0.00	0.01	0.00	0.01
Al ₂ O ₃	33.57	34.59	33.96	33.41	33.73	33.50	32.90	33.34	34.10	33.65	33.36	32.85	33.61	34.54
FeO	3.96	4.05	4.21	4.07	3.86	4.38	3.86	4.75	4.81	5.17	3.56	3.68	3.85	3.61
MnO	0.15	0.08	0.17	0.13	0.09	0.17	0.10	0.24	0.15	0.12	0.22	0.17	n.a.	n.a.
MgO	10.84	11.25	11.14	11.01	11.07	10.57	10.85	10.29	10.24	10.06	10.95	10.99	11.08	11.02
CaO	0.01	0.01	0.00	0.01	0.02	0.03	0.02	0.00	0.01	0.02	0.01	0.01	0.01	0.03
Na ₂ O	0.20	0.19	0.24	0.22	0.21	0.23	0.22	0.27	0.32	0.21	0.39	0.54	0.22	0.19
K ₂ O	0.01	0.01	0.01	0.03	0.00	0.02	0.02	0.00	0.01	0.03	0.03	0.02	0.02	0.02
Cr ₂ O ₃	0.01	0.02	0.00	0.01	0.00	0.00	0.00	0.00	0.01	0.00	0.02	0.00	n.a.	n.a.
Total	97.89	99.69	98.90	97.93	98.13	96.86	96.66	97.11	98.47	97.16	98.04	96.19	98.94	100.05
Si	4.986	4.933	4.947	4.979	4.971	4.933	5.000	4.955	4.945	4.927	5.010	4.957	5.024	5.007
Ti	0.001	0.000	0.001	0.000	0.000	0.001	0.002	0.000	0.001	0.001	0.000	0.001	0.000	0.001
Al	4.012	4.064	4.027	3.996	4.021	4.062	3.981	4.038	4.072	4.080	3.977	4.003	3.968	4.026
Fe ²⁺	0.336	0.338	0.354	0.345	0.327	0.377	0.331	0.408	0.408	0.445	0.301	0.318	0.323	0.298
Mn	0.013	0.007	0.014	0.011	0.008	0.015	0.009	0.021	0.013	0.010	0.018	0.015	n.a.	n.a.
Mg	1.638	1.672	1.671	1.666	1.669	1.621	1.661	1.576	1.547	1.543	1.652	1.693	1.655	1.625
Ca	0.001	0.001	0.000	0.001	0.002	0.003	0.002	0.000	0.001	0.002	0.001	0.001	0.001	0.003
Na	0.038	0.037	0.047	0.043	0.041	0.046	0.044	0.054	0.063	0.042	0.077	0.107	0.042	0.036
K	0.001	0.001	0.001	0.003	0.000	0.003	0.003	0.000	0.001	0.004	0.004	0.002	0.003	0.003
Cr	0.000	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.001	0.000	n.a.	n.a.
Total	11.027	11.053	11.063	11.046	11.039	11.060	11.031	11.053	11.050	11.055	11.041	11.096	11.015	10.998
Al(T2)	0.014	0.067	0.053	0.021	0.029	0.067	0.000	0.045	0.055	0.073	0.000	0.043	0.000	0.000
AI(T1)	3.998	3.997	3.974	3.975	3.992	3.995	3.981	3.993	3.998	3.999	3.977	3.959	3.968	3.999
Fe(T1)	0.001	0.002	0.025	0.024	0.008	0.005	0.018	0.007	0.000	0.000	0.021	0.040	0.032	0.000
Al(M)	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.018	0.008	0.000	0.000	0.000	0.026
Fe(M)	0.335	0.336	0.329	0.321	0.319	0.372	0.314	0.402	0.408	0.445	0.279	0.278	0.291	0.298
X _{Mg}	0.824	0.829	0.820	0.824	0.833	0.805	0.830	0.786	0.786	0.772	0.838	0.836	0.837	0.845

Appendix 3. Chemical compositions of cordierites.

n = number of analyses

n.a. = not analysed

mineral formula based on 36 charges or 18 oxygens

Assemblages:

Oam-Crd-Ky rock (zone)

Oam-Crd-Ky FOCK (2011) 1. Crd-Ky-Qtz-Pl-Bt-Chl-Ath 2. Crd-Chl-Ged-Ath-Qtz-Bt 3. Crd-Bt-Ath 4. Crd-Ath-Pl-Chl-Qtz 6. Crd-Bt-Qtz 7. Crd-Ath-Qtz

Ky-St-Crd rock (zone) 8. Crd-St-Ky-Sil-Pl-Qtz 9. Crd-St-Bt-Ilm 10. Crd-St-Pl-Qtz-Bt

5. Crd-Ath-Bt-Qtz-Chl-Pl Ky-Chl-Crd rock (coarse-grained) 11. Crd-Ky-Bt-Chl-Pl-Qtz

12. Crd-Ky-Bt-Chl

Ky-Crd-Sil rock (zone) 13. Crd-Ky-Sil-Bt-Ms-Pl Qtz 14. Crd-Ky-Sil-Bt-Ms-Pl-Qtz

Appendix 4. Chemical compositions of staurolites.

Appendix 5. Chemical compositions of garnets.

Analysis no.	15	16	17	18	Analysis no.	19	20	21	22	23	24	25
Occurrence	Hiltuspuro:				Occurrence	Hiltusp	aro:					
Sample/ring	10A/1	10A/1	10A/2	10C	Sample/ring	48B1/1	48B1/1	48B1/1	48B1/1	48B1/6	48B1/6	48B2/1
n	1	1	2	1								
SIO	27 79	27 36	27 50	27.28	SiO ₂	36.83	36.90	36.67	36.94	36.40	36.13	35.91
TIO	0.43	0.37	0.62	0.59	TiO ₂	0.00	0.01	0.00	0.03	0.01	0.04	0.01
ALO	54 31	53 46	52.85	52.25	Al ₂ O ₃	20.63	20.58	20.35	20.53	20.70	20.34	20.59
FeO	12.89	12 13	13.16	13 32	FeO	27.39	26.88	27.08	27.18	35.31	33.51	36.27
MnO	0.33	0.31	0.30	0.26	MnO	9.93	9.42	9.56	10.00	5.05	5.63	4.17
MgO	2.55	2.71	2.92	2.87	MgO	1.83	2.17	1.80	1.87	0.90	1.54	1.17
CaO	0.00	0.00	0.01	0.01	CaO	3.01	3.82	3.48	3.43	1.13	1.19	1.14
Na ₂ O	0.00	0.04	0.02	0.02	CroO	0.04	0.03	0.04	0.02	0.00	0.00	0.00
K ₂ O	0.00	0.01	0.01	0.03	Total	99.66	09 80	98.97	100.00	99.49	98 37	99.25
Cr ₂ O ₃	0.13	0.02	0.05	0.01	Total	77.00	77.00	10.71	100.00	JJ. 4J	20.57	11.40
Total	98.43	96.38	97.43	96.62	Si	2.997	2.992	3.003	2.997	2.997	2.998	2.971
C!	2 005	2014	2016	2 001	Ti	0.000	0.000	0.000	0.002	0.000	0.002	0.000
SI	3.905	3.914	3.915	3.921	Al	1.979	1.966	1.965	1.963	2.008	1.989	2.008
11	0.045	0.039	0.067	0.004	Fe ²⁺	1.864	1.823	1.855	1.845	2,431	2.325	2.510
AI E-2+	8.993	9.013	1.567	1 601	Mn	0.685	0.647	0.663	0.687	0 352	0 395	0 292
re	1.515	0.039	1.307	0.032	Ma	0.222	0.263	0.219	0.226	0.110	0 191	0 144
Ma	0.534	0.577	0.630	0.615	Ca	0.262	0.220	0.205	0.200	0.100	0.106	0.101
Co	0.000	0.000	0.001	0.001	Ca	0.203	0.002	0.303	0.298	0.100	0.100	0.101
Na	0.000	0.011	0.001	0.001	Cr	0.002	0.002	0.002	0.001	0.000	0.000	0.000
K	0.000	0.002	0.002	0.005	Total	8.012	8.024	8.013	8.019	7.999	8.006	8.025
Cr	0.014	0.002	0.005	0.001	Alma	0 615	0 505	0 610	0.604	0.912	0 771	0.824
Total	15.046	15.046	15.086	15.094	Am	0.015	0.333	0.010	0.004	0.012	0.121	0.024
					Sps	0.220	0.211	0.218	0.225	0.110	0.151	0.090
AI(T)	0.095	0.086	0.085	0.079	Prp	0.073	0.086	0.072	0.074	0.037	0.003	0.047
AIMO	8,898	8.927	8.782	8.772	Grs	0.087	0.108	0.100	0.098	0.033	0.035	0.033
AI(M4+M6)	0.000	0.000	0.000	0.000		mineral	formula b	ased on 2	4 charges	or 12 oxy	gens	
Fe(M6)	0.087	0.071	0 212	0 227								
Fe(M4+M6)	1 427	1 379	1 355	1 374		Assemb	lages:					
Y.	0.256	0.279	0 279	0 274								
Mg	0.200	0.217	0.217	0.274		Grt-Qtz	rock					

n = number of analyses

mineral formula based on 47 charges or 23.5 oxygens

Assemblages:

Ky-St-Crd rock (zone)

15. St-Crd-Ky-Sil-Pl-Qtz 16. St-Crd-Ky-Sil-Pl-Qtz 17. St-Crd-Bt-Ilm 18. St-Crd-Pl-Qtz-Bt
 19. rim, Grt-Bt-Chl-Qtz
 23.

 20. core, Grt-Bt-Chl-Qtz
 24.

 21. rim, Grt-Bt-Chl-Qtz
 25.

 22. rim, Grt-Bt-Chl-Qtz
 25.

.

23. rim, Grt-Ms-Pl-Qts 24. core, Grt-Ms-Pl-Qts

25. core, Grt-Ms-Pl-Qtz

Analysis no.	26	27	28	29	30	31	32	33	34
Occurrence	Hiltuspur	0:		1					
Sample/ring	1A1/1	1A1/2	1A1/2	1A1/2	1A1/2	1A1/3	1A2/3	1A2/3	1A2/3
SiO ₂	53.47	46.00	53.99	55.33	56.05	54.43	54.88	54.60	54.88
TiO ₂	0.03	0.26	0.04	0.06	0.03	0.04	0.00	0.03	0.02
Al ₂ O ₃	3.69	13.43	3.25	2.88	2.38	3.00	2.67	2.18	1.43
FeO	18.57	19.36	19.47	18.11	19.02	19.55	19.58	17.76	17.73
MnO	0.65	0.61	0.60	0.59	0.59	0.54	0.59	0.62	0.69
MgO	21.95	16.74	20.11	22.01	22.33	22.03	22.15	21.81	22.08
CaO	0.13	0.13	0.07	0.10	0.06	0.08	0.09	0.09	0.04
Na ₂ O	0.21	1.55	0.25	0.23	0.16	0.17	0.22	0.19	0.06
K ₂ O	0.00	0.02	0.02	0.01	0.00	0.01	0.01	0.01	0.00
Total	98.70	98.10	97.80	99.32	100.62	99.85	100.19	97.29	96.93
Si	7.561	6.635	7.718	7.730	7.754	7.627	7.664	7.787	7.852
Ti	0.003	0.028	0.004	0.006	0.003	0.004	0.000	0.003	0.002
Al	0.615	2.283	0.548	0.474	0.388	0.495	0.439	0.366	0.241
Fe ²⁺	2.196	2.335	2.328	2.116	2.200	2.291	2.287	2.118	2.122
Mn	0.078	0.075	0.073	0.070	0.069	0.064	0.070	0.075	0.084
Mg	4.627	3.600	4.286	4.584	4.605	4.602	4.611	4.637	4.710
Ca	0.020	0.020	0.011	0.015	0.009	0.012	0.013	0.014	0.006
Na	0.058	0.433	0.069	0.062	0.043	0.046	0.060	0.053	0.017
K	0.000	0.004	0.004	0.002	0.000	0.002	0.002	0.002	0.000
Total	15.157	15.414	15.040	15.059	15.071	15.145	15.147	15.054	15.033
Al(T1)	0.439	1.365	0.282	0.270	0.246	0.373	0.336	0.213	0.148
AI(M2)	0.176	0.918	0.266	0.204	0.142	0.123	0.104	0.153	0.093
Ti(M2)	0.003	0.028	0.004	0.006	0.003	0.004	0.000	0.003	0.002
Fe(M2)	0.586	0.414	0.609	0.565	0.600	0.622	0.629	0.578	0.591
Mg(M2)	1.235	0.639	1.121	1.225	1.255	1.250	1.268	1.266	1.313
Fe(M1+M3)	0.966	1.180	1.056	0.947	0.970	0.997	0.995	0.941	0.932
Mg(M1+M3)	2.034	1.820	1.944	2.053	2.030	2.003	2.005	2.059	2.068
Fe(M4)	0.644	0.740	0.663	0.603	0.630	0.672	0.664	0.599	0.598
Mg(M4)	1.358	1.141	1.221	1.307	1.319	1.349	1.338	1.312	1.328
Mn(M4)	0.078	0.075	0.073	0.070	0.069	0.064	0.070	0.075	0.084
Ca(M4)	0.020	0.020	0.011	0.015	0.009	0.012	0.013	0.014	0.006
Na(M4)	0.000	0.023	0.031	0.005	0.000	0.000	0.000	0.000	0.000
K(M4)	0.000	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.000
Na(A)	0.058	0.410	0.038	0.057	0.043	0.046	0.060	0.052	0.017
K(A)	0.000	0.003	0.003	0.002	0.000	0.002	0.002	0.002	0.000
Vac(A)	0.942	0.586	0.959	0.941	0.957	0.952	0.939	0.946	0.983
X _{Mg}	0.678	0.607	0.648	0.684	0.677	0.668	0.668	0.686	0.689

Appendix 6. Chemical compositions of orthoamphiboles.

mineral formula based on 46 charges or 23 oxygens; 15 cations, all Fe as Fe²⁺.

Assemblages:

Oam-Crd-Ky rock (zone)

Ath-Crd-Ky-Qtz-Pl-Bt-Chl
 Ged-Ath-Crd-Chl-Qtz-Bt
 Ath-Ged-Crd-Chl-Qtz-Bt
 Ath-Crd-Bt
 Ath-Crd-Bt

Ath-Crd-Pl-Chl-Qtz
 Ath-Crd-Qtz
 Ath-Crd-Qtz
 Ath-Crd-Qtz
 Ath-Crd-Qtz

Analysis no.	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49
Occurrence	Havukk	alammit	:				Hiltuspu	Iro:							
Sample/ring	25A1/2	25B/1	25B/2	25B/3	25B/3	25B/5	1A1/1	1A1/3	1A2/1	3	10A/1	10C	12A	48B1/6	48B2/1
n	1	1	1	1	1	1	1	2	3	2	3	2	2	1	1
SiO	54.14	46.71	46.76	54.02	51.60	49.82	67.95	66.21	65.78	65.56	66.20	65.34	67.11	65.98	63.81
TiO	0.02	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.04	0.00	0.02	0.01	0.00	0.00	0.00
AbO	29.68	34.73	35.14	30.41	30.02	33.27	20.56	21.91	20.33	21.32	20.69	21.87	20.69	22.27	23.06
FeO	0.03	0.02	0.06	0.04	0.03	0.05	0.00	0.04	0.21	0.01	0.03	0.05	0.02	0.09	0.07
MnO	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a	n.a	n.a	n.a	n.a	n.a	n.a.	0.00	n.a.
MgO	0.03	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.01	0.00	0.02	0.02	0.01	0.00	0.00
CaO	11.20	15.51	17.37	11.95	12.83	14.65	0.83	2.06	0.45	2.29	1.69	1.87	1.30	2.09	3.56
Na ₂ O	4.65	1.77	1.56	4.54	4.13	2.70	11.07	9.74	11.22	10.67	10.95	10.43	10.76	9.94	9.40
K ₂ O	0.03	0.01	0.03	0.03	0.06	0.00	0.03	0.01	0.06	0.04	0.03	0.10	0.03	0.05	0.07
Total	99.78	98.75	100.93	100.99	98.67	100.49	100.45	99.97	98.10	99.89	99.64	99.66	99.91	100.42	99.97
Si	2.442	2.158	2.127	2.414	2.371	2.254	2.956	2.897	2.937	2.887	2.918	2.879	2.939	2.879	2.814
Ti	0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.001	0.000	0.000	0.000	0.000
Al	1.578	1.891	1.884	1.601	1.626	1.774	1.054	1.130	1.070	1.107	1.075	1.135	1.068	1.146	1.198
Fe ²⁺	0.001	0.001	0.002	0.001	0.001	0.002	0.000	0.001	0.008	0.000	0.001	0.002	0.001	0.003	0.003
Mn	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	0.000	n.a.
Mg	0.002	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.001	0.000	0.001	0.001	0.001	0.000	0.000
Ca	0.541	0.768	0.847	0.572	0.632	0.710	0.039	0.097	0.022	0.108	0.080	0.088	0.061	0.098	0.168
Na	0.407	0.159	0.138	0.393	0.368	0.237	0.934	0.826	0.971	0.911	0.935	0.891	0.913	0.841	0.804
K	0.002	0.001	0.002	0.002	0.004	0.000	0.002	0.001	0.003	0.002	0.002	0.005	0.002	0.003	0.004
Total	4.973	4.976	5.000	4.983	5.002	4.977	4.985	4.952	5.014	5.016	5.013	5.001	4.984	4.970	4.991
An	0.570	0.828	0.859	0.592	0.630	0.750	0.040	0.105	0.022	0.106	0.079	0.089	0.063	0.104	0.172
Ab	0.428	0.171	0.140	0.407	0.367	0.250	0.959	0.895	0.975	0.892	0.920	0.905	0.936	0.893	0.824

Appendix 7. Chemical compositions of plagioclases.

0.002

Or

n = number of analyses

0.001

n.a. = not analysed

mineral formula based on 16 charges or 8 oxygens

0.002

0.002

0.004

Assemblages:

Ky-Bt-Pl rock (coarsed grained)

Pl-Bt-Ky (earlier phase)
 Pl-Bt (earlier phase)
 Pl-Bt (earlier phase)

Pl-Bt-Ky (later phase)
 Pl-Bt-Ky (later phase)
 Pl-Bt-Ky (later phase)

Oam-Crd-Ky rock (zone) 41. Pl-Crd-Ky-Ath-Bt-Chl-Qtz 42. Pl-Crd-Ath-Chl-Qtz 43. Pl-Cdr-Ath-Bt-Chl-Qtz

Leucotonalite

44. Pl-Bt-Qtz

0.000

0.002

0.001

Ky-St-Crd rock (zone) 45. Pl-Crd-St-Ky-Sil-Qtz 46. Pl-Crd-St-Bt-Qtz

0.003

Ky-Chl-Crd rock (coarse grained) 47. Pl-Crd-Ky-Bt-Chl-Qtz

0.002

0.002 0.005

Grt-Qtz rock 48. Pl-Grt-Ms-Qtz 49. Pl-Grt-Ms-Qtz

0.002

0.003 0.004

Ky-Crd-Sil rock (zone) 50. Pl-Crd-Ky-Sil-Bt-Ms-Qtz 50 Tetrilampi:

15B2/1

64.89

0.03

19.48

0.00

n.a.

0.00

0.51

11.59

0.04

96.54

2.948

1 043

0.000

0.000

0.025

1.021

0.002

5.041

0.024

0.974

0.002

n.a.

Analysis no.	51	52	53	54	55	56	57	58	59	60	61	62
Occurrence	Havukk	alammit	:						Tetrilan	npi:		
Sample/ring	25A1/1	25A1/2	25B/1	25B/2	25B/3	25B/3	25B/4	25B/5	15B2/1	15B2/1	15B2/2	15B2/2
n	1	1	1	1	1	1	1	1	1	1	1	1
SiO ₂	38.73	37.31	36.38	36.56	37 34	37 31	36 97	37.06	38 16	38 43	38 29	37 72
TiO ₂	0.58	0.53	0.59	0.52	0.47	0.64	0.96	0.56	1.21	1.08	1 10	1 12
AbO	20.77	19.19	20.77	19.65	20.95	20.91	20 33	20.62	20.23	20.57	20.47	21 21
FeO	11.66	11.98	13.76	13.28	12.33	12.78	13.23	13.16	10.82	10 79	10 51	11.02
MnO	0.11	0.16	0.20	0.17	0.13	0.15	0.14	0.15	0.04	0.13	0.07	0.14
MgO	17.11	17.20	15.60	15.29	15.49	15.31	15.79	15.68	17.03	17.21	17.43	16 44
CaO	0.04	0.00	0.15	0.04	0.01	0.03	0.03	0.05	0.00	0.00	0.00	0.03
Na ₂ O	0.34	0.21	0.27	0.31	0.47	0.39	0.22	0.37	0.32	0.36	0.37	0.30
K ₂ O	8.80	9.02	8.38	8.95	8.62	9.16	9.32	8.72	8.82	8 70	8 53	8 61
Total	98.15	95.58	96.10	94.76	95.81	96.67	96.98	96.38	96.63	97.28	96.77	96.60
Si	2.724	2.717	2.649	2,704	2.704	2.692	2.671	2.684	2,720	2717	2716	2 689
Ti	0.030	0.029	0.032	0.029	0.026	0.035	0.052	0.031	0.065	0.058	0.059	0.060
Al	1.722	1.646	1.783	1.713	1.788	1.778	1.731	1.760	1.700	1.715	1.711	1 782
Fe ²⁺	0.686	0.729	0.838	0.822	0.747	0.771	0.799	0.797	0.645	0.638	0.624	0.657
Mn	0.007	0.010	0.012	0.010	0.008	0.009	0.008	0.009	0.003	0.008	0.004	0.008
Mg	1.794	1.866	1.693	1.685	1.672	1.647	1.701	1.693	1.810	1.815	1.843	1.748
Ca	0.003	0.000	0.011	0.003	0.001	0.002	0.002	0.004	0.000	0.000	0.000	0.003
Na	0.047	0.029	0.039	0.044	0.066	0.055	0.031	0.052	0.044	0.050	0.050	0.042
K	0.790	0.838	0.778	0.844	0.797	0.843	0.859	0.806	0.802	0.785	0.772	0.783
Total	7.803	7.864	7.836	7.855	7.808	7.832	7.856	7.834	7.788	7.785	7.781	7.772
Si(T2)	0.724	0.717	0.649	0.704	0.704	0.692	0.671	0.684	0 720	0 717	0 716	0 689
AI(T2)	1.276	1.283	1.351	1.296	1.296	1.308	1.329	1.316	1.280	1 283	1 284	1 311
AI(M2)	0.446	0.363	0.432	0.417	0.492	0.471	0 403	0 444	0.420	0 432	0.428	0 472
Fe(M2)	0.420	0.449	0.504	0.506	0.455	0.474	0 491	0.485	0 397	0 391	0 382	0 399
Mg(M2)	1.097	1.149	1.019	1.038	1.019	1.012	1.046	1.031	1.115	1 112	1 128	1.061
Fe(M1)	0.266	0.280	0.333	0.316	0.291	0.297	0.308	0.311	0.247	0.247	0.242	0.258
Mg(M1)	0.697	0.717	0.674	0.648	0.653	0.635	0.656	0.662	0.695	0.703	0.716	0.686
Vac(A)	0.257	0.191	0.261	0.200	0.269	0.212	0.172	0.246	0.242	0.265	0.278	0.259
												0.201

Appendix 8. Chemical compositions of biotites.

n = number of analyses

mineral formula based on 22 charges or 11 oxygens

Assemblages:

Ky-Bt-Pl rock (coarse grained)

51. Bt-Pl-Ky-Ms-Chl 52. Bt-Pl-Ky 53. Bt-Pl 54. Bt-Pl 55. Bt-Pl-Ky 56. Bt-Pl-Ky 57. Bt-Chl-Ky-Pl 58. Bt-Pl-Ky

Ky-Crd-Sil rock (zone)

59. Bt-Crd-Ky-Sil-Ms-Pl-Qtz
 60. Bt-Crd-Ky-Sil-Ms-Pl-Qtz
 61. Bt-Crd-Ky-Sil-Ms-Pl-Qtz
 62. Bt-Crd-Ky-Sil-Ms-Pl-Qtz

Append	ix 8.	cont.
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Analysis no.	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79
Occurrence	Hiltusp	uro:															
Sample/ring	1A1/1	1A1/2	1A1/2	1A2/2	1A2/4	1A2/4	3	10A/1	10A/2	10A/2	10C	10C	12A	12A	12C1	48B1/1	48B1/1
n	1	1	2	2	1	1	1	2	1	2	2	1	2	1	4	1	1
SiO ₂	39.02	39.47	39.35	38.26	38.55	38.50	37.53	36.70	37.40	37.39	36.88	35.95	37.93	38.48	38.45	35.65	35.29
TiO ₂	1.17	1.19	1.24	1.13	1.06	1.04	1.12	1.08	1.17	0.99	1.25	1.04	1.32	0.96	1.20	1.13	0.90
Al ₂ O ₃	18.31	17.69	17.06	17.13	18.20	17.45	19.25	18.97	18.32	18.35	18.59	18.96	18.06	19.09	19.34	19.20	19.24
FeO	12.04	12.70	11.43	11.82	12.13	12.47	13.32	14.93	16.28	15.15	16.58	16.03	11.68	11.01	9.58	24.81	22.30
MnO	0.01	0.03	0.04	0.04	0.02	0.00	0.07	0.06	0.09	0.09	0.04	0.03	0.13	0.09	0.07	0.81	0.78
MgO	16.68	17.83	16.98	17.91	17.42	17.52	15.53	14.63	14.47	14.46	14.13	13.59	17.64	17.43	16.66	6.68	7.13
CaO	0.02	0.02	0.04	0.02	0.00	0.00	0.01	0.17	0.01	0.00	0.01	0.03	0.02	0.01	0.01	0.00	0.01
Na ₂ O	0.37	0.29	0.44	0.48	0.75	0.54	0.28	0.46	0.45	0.39	0.21	0.29	0.34	0.38	0.35	0.13	0.10
K ₂ O	8.32	8.37	8.29	8.32	8.18	8.21	9.22	8.28	8.66	8.81	9.08	8.89	8.65	8.44	9.29	9.09	9.25
Total	95.94	97.59	94.85	95.08	96.31	95.73	96.33	95.25	96.85	95.62	96.76	94.81	95.74	95.89	94.94	97.52	94.99
Si	2.809	2.803	2.859	2.789	2.773	2.791	2.729	2.713	2.739	2.760	2.714	2.696	2.749	2.763	2.782	2.718	2.736
Ti	0.063	0.064	0.068	0.062	0.057	0.057	0.061	0.060	0.064	0.055	0.069	0.059	0.072	0.052	0.065	0.065	0.052
Al	1.554	1.481	1.461	1.472	1.543	1.491	1.650	1.653	1.581	1.596	1.613	1.676	1.542	1.616	1.648	1.725	1.758
Fe ²⁺	0.725	0.754	0.695	0.721	0.730	0.756	0.810	0.923	0.997	0.935	1.021	1.005	0.708	0.661	0.579	1.582	1.446
Mn	0.001	0.002	0.002	0.002	0.001	0.000	0.004	0.003	0.006	0.006	0.002	0.002	0.008	0.005	0.004	0.052	0.051
Mg	1.790	1.888	1.840	1.946	1.868	1.894	1.683	1.612	1.580	1.591	1.550	1.520	1.905	1.866	1.797	0.759	0.824
Ca	0.002	0.002	0.003	0.001	0.000	0.000	0.001	0.013	0.001	0.000	0.001	0.002	0.001	0.001	0.001	0.000	0.001
Na	0.052	0.040	0.061	0.068	0.105	0.076	0.039	0.066	0.064	0.056	0.030	0.042	0.047	0.053	0.049	0.020	0.015
K	0.764	0.758	0.769	0.773	0.751	0.759	0.855	0.780	0.809	0.829	0.852	0.851	0.800	0.773	0.857	0.884	0.915
Total	7.759	7.792	7.757	7.834	7.826	7.824	7.833	7.824	7.842	7.829	7.852	7.853	7.832	7.790	7.782	7.806	7.798
Si(T2)	0.809	0.803	0 859	0.789	0.773	0.791	0.729	0.713	0.739	0.760	0.714	0.696	0.749	0.763	0.782	0.718	0.736
AI(T2)	1 191	1 197	1 141	1.211	1.227	1.209	1.271	1.287	1.261	1.240	1.286	1.304	1.251	1.237	1.218	1.282	1.264
AI(M2)	0.363	0.284	0.320	0.261	0.315	0.282	0.378	0.366	0.321	0.357	0.327	0.372	0.291	0.379	0.430	0.443	0.494
Fe(M2)	0.454	0.471	0.441	0.453	0.457	0.474	0.506	0.572	0.623	0.586	0.636	0.624	0.441	0.409	0.366	0.972	0.893
Ma(M2)	1 120	1 179	1 168	1.222	1.169	1,187	1.051	0.999	0.986	0.997	0.966	0.943	1.188	1.155	1.135	0.467	0.510
Fe(M1)	0 271	0 283	0 253	0.268	0.273	0.282	0.304	0.351	0.375	0.349	0.385	0.382	0.267	0.252	0.213	0.610	0.552
Mg(M1)	0.670	0.709	0.671	0.724	0.698	0.707	0.633	0.613	0.594	0.595	0.584	0.577	0.717	0.711	0.662	0.293	0.315
Vac(A)	0 288	0 282	0 292	0 295	0.354	0.317	0.184	0.286	0.255	0.227	0.178	0.191	0.247	0.280	0.192	0.136	0.100
Y	0.200	0 714	0.725	0 729	0 719	0.715	0.674	0.635	0.612	0.628	0.602	0.602	0.727	0.737	0.755	0.317	0.355
A Mg	n = nun	ber of a	nalvses	0.727	0.715	0.710	0.071	0.000									

mineral formula based on 22 charges or 11 oxygens

Assemblages:

Oam-Crd-Ky rock (zone) 63. Bt-Crd-Ky-Chl-Ath-Pl-Qtz 64. Bt-Crd-Ath 65. Bt-Crd-Ath 66. Bt-Crd-Ath 67. Bt-Pl-Crd 68. Bt-Pl-Crd

Leucotonalite 69. Bt-Pl-Qtz Ky-St-Crd rock (zone) 70. Bt-Crd-St-Ky-Sil-Pl-Qtz 71. Bt-Crd-St-Ilm 72. Bt-Crd-St-Ilm 73. Bt-Crd-St-Pl-Qtz 74. Bt-Crd-St-Pl-Qtz Grt-Qtz rock 78. Bt-Grt-Chl-Qtz 79. Bt-Grt-Chl-Qtz

Ky-Chl-Crd rock (coarse grained) 75. Bt-Crd-Ky-Chl-Pl-Qtz 76. Bt-Crd-Ky-Chl-Pl-Qtz 77. Bt-Crd-Ky-Chl

Analysis no.	80	81	82	83	84	85	86	87	88	89	90	91	92	93
Occurrence	Havuk	alammi	t:	Hiltusp	uro:									
Sample/ring	25A1/1	25A1/1	25B/4	1A1/1	1A1/1	1A1/2	1A1/2	1A1/3	12A	12A	12C1	48B1/1	48B1/1	48B1/1
n	1	1	1	1	1	2	2	1	3	2	2	1	1	1
SiO ₂	26.71	26.47	25.78	26.14	26.93	27.43	27 13	27 31	25 76	26 40	27 40	24.85	23.97	24 52
TiO ₂	0.00	0.01	0.02	0.05	0.07	0.10	0.06	0.04	0.07	0.06	0.09	0.08	0.04	0.01
Al ₂ O ₃	25.97	26.15	23.50	23.59	23.22	22.87	23.90	22.58	23.55	23 74	23 74	22.05	23 31	23 19
FeO	14.04	16.32	15.14	13.14	13.09	12.48	12.84	13.31	13.67	13.88	10.91	31.37	31 58	32.78
MnO	0.18	0.28	0.22	0.03	0.09	0.05	0.06	0.08	0.14	0.10	0.18	0.59	0.87	0.99
MgO	23.99	21.46	22.83	24.01	23.70	23.22	23.78	22.28	24.94	24.42	24 67	9.86	9 31	10 39
CaO	0.01	0.06	0.02	0.00	0.02	0.01	0.02	0.03	0.01	0.02	0.01	0.00	0.00	0.00
Na ₂ O	0.02	0.01	0.00	0.03	0.03	0.02	0.01	0.03	0.04	0.02	0.03	0.05	0.11	0.02
K ₂ O	0.04	0.04	0.03	0.03	0.01	0.08	0.04	0.04	0.02	0.03	0.04	0.02	0.13	0.02
Total	90.95	90.81	87.54	87.02	87.16	86.26	87.83	85.70	88.19	88.65	87.04	88.87	89.32	91.91
Si	2.543	2.553	2.577	2.597	2.666	2.731	2.658	2.749	2.539	2.584	2.682	2.670	2.573	2.565
Ti	0.000	0.001	0.002	0.004	0.005	0.007	0.004	0.003	0.005	0.004	0.007	0.006	0.003	0.001
Al	2.914	2.972	2.769	2.762	2.709	2.684	2.760	2.678	2.735	2.739	2.739	2.792	2.949	2.859
Fe ²⁺	1.118	1.316	1.266	1.092	1.084	1.039	1.052	1.120	1.127	1.136	0.893	2.819	2.835	2.868
Mn	0.015	0.023	0.019	0.003	0.008	0.004	0.005	0.007	0.012	0.008	0.015	0.054	0.079	0.088
Mg	3.405	3.085	3.402	3.556	3.497	3.446	3.473	3.343	3.664	3.564	3.600	1.579	1.490	1.620
Ca	0.001	0.006	0.002	0.000	0.002	0.001	0.002	0.003	0.001	0.002	0.001	0.000	0.000	0.000
Na	0.004	0.002	0.000	0.006	0.006	0.004	0.002	0.006	0.008	0.004	0.006	0.010	0.023	0.004
K	0.005	0.005	0.004	0.004	0.001	0.010	0.005	0.005	0.003	0.004	0.005	0.003	0.018	0.003
Total	10.004	9.964	10.039	10.023	9.978	9.927	9.961	9.915	10.093	10.046	9.947	9.934	9.970	10.008
Si(T2)	0.543	0.553	0.577	0.597	0.666	0.731	0.658	0.749	0.539	0.584	0.682	0.670	0.573	0.565
AI(T2)	1.457	1.446	1.422	1.399	1.329	1.262	1.338	1.248	1.456	1.411	1.311	1.323	1.424	1 434
AI(M2)	1.457	1.526	1.347	1.363	1.380	1.422	1.422	1.430	1.279	1.328	1.427	1.469	1.525	1.425
Fe(M2)	0.134	0.141	0.176	0.150	0.146	0.134	0.134	0.143	0.169	0.162	0.113	0.336	0.306	0.360
Mg(M2)	0.407	0.331	0.474	0.487	0.472	0.444	0.443	0.426	0.550	0.509	0.457	0.188	0.161	0.204
Mn(M2)	0.002	0.002	0.003	0.000	0.001	0.001	0.001	0.001	0.002	0.001	0.002	0.006	0.009	0.011
Fe(M1)	0.984	1.175	1.089	0.942	0.937	0.905	0.918	0.978	0.958	0.974	0.780	2.483	2.529	2.508
Mg(M1)	2.998	2.755	2.928	3.069	3.025	3.003	3.030	2.917	3.115	3.055	3.143	1.391	1.329	1.417
Mn(M1)	0.013	0.020	0.016	0.002	0.007	0.004	0.004	0.006	0.010	0.007	0.013	0.047	0.071	0.077
X _{Mg}	0.750	0.697	0.726	0.765	0.762	0.768	0.767	0.748	0.763	0.757	0.799	0.355	0.338	0.354
	n = numl	per of an	alvses											

Appendix 9. Chemical compositions of chlorites.

mineral formula based on 28 charges or 14 oxygens

Assemblages:

Ky-Bt-Pl rock (coarse grained)

80. Chl-Ky-Bt-Ms-Pl 81. Chl-Ky-Bt-Ms-Pl 82. Chl-Bt-Ky-Pl

Oam-Crd-Ky rock (zone)

Chl-Crd-Ky-Bt-Ath-Pl-Qtz
 Chl-Crd-Ky-Bt-Ath-Pl-Qtz
 Chl-Crd-Ged-Ath-Qtz-Bt
 Chl-Crd-Ged-Ath-Qtz-Bt
 Chl-Crd-Ged-Ath-Pl-Qtz

Ky-Chl-Crd rock (coarse grained)

88. Chl-Crd-Ky-Bt-Pl-Qtz
 89. Chl-Crd-Ky-Bt-Pl-Qtz
 90. Chl-Crd-Ky-Bt

Grt-Qtz rock

91. Chl-Grt-Bt-Qtz 92. Chl-Grt-Bt-Qtz 93. Chl-Grt-Bt-Qtz

Analysis no.	94	95	96	97	98	99	100	101	102	103
Occurrence	Havukka	lammit:	Hiltuspu	ro:		Tetrilam	pi:			
Sample/ring	25A1/1	25A1/1	1A2/4	1A2/4	48B1/6	15B2/1	15B2/1	15B2/1	15B2/2	15B2/2
n	1	1	1	1	8	1	1	1	1	1
SiO ₂	45.38	45.91	47.33	48.98	46.36	45.60	46.71	45.05	46.74	46.82
TiO ₂	0.05	0.18	0.00	0.01	0.19	0.48	0.61	0.48	0.70	0.63
Al ₂ O ₃	35.95	36.86	37.59	32.42	37.91	35.80	35.90	36.66	35.61	36.16
FeO	2.18	2.36	0.85	1.60	1.21	1.73	2.06	2.00	2.00	1.91
MnO	0.00	0.00	0.04	0.00	0.00	0.00	0.00	0.00	0.00	0.00
MgO	0.45	0.38	1.50	6.00	0.28	0.55	0.58	0.55	0.68	0.78
CaO	0.05	0.06	0.04	0.04	0.01	0.02	0.00	0.02	0.01	0.00
Na ₂ O	1.31	1.34	6.02	4.49	0.92	1.83	1.80	2.14	1.90	1.65
K ₂ O	8.85	8.40	1.26	1.20	9.62	7.95	8.09	7.97	7.75	7.46
Total	94.41	95.49	94.63	94.74	96.49	94.16	95.85	95.02	95.68	95.75
Si	3.046	3.033	3.049	3.156	3.027	3.052	3.071	2.998	3.079	3.072
Ti	0.002	0.009	0.000	0.000	0.009	0.024	0.030	0.024	0.035	0.031
Al	2.845	2.870	2.854	2.462	2.917	2.824	2.782	2.875	2.765	2.797
Fe ²⁺	0.122	0.130	0.046	0.086	0.066	0.097	0.113	0.111	0.110	0.105
Mn	0.000	0.000	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Mg	0.045	0.037	0.144	0.576	0.027	0.055	0.057	0.055	0.067	0.077
Ca	0.004	0.004	0.003	0.003	0.001	0.002	0.000	0.001	0.001	0.000
Na	0.171	0.171	0.752	0.561	0.117	0.237	0.230	0.276	0.243	0.210
к	0.758	0.708	0.104	0.099	0.801	0.679	0.679	0.677	0.651	0.624
Total	6.993	6.963	6.952	6.943	6.964	6.970	6.962	7.017	6.951	6.916
Si(T2)	1.046	1.033	1.049	1.156	1.027	1.052	1.071	0.998	1.079	1.072
AI(T2)	0.954	0.967	0.951	0.844	0.973	0.948	0.929	1.002	0.921	0.928
AI(M2)	1.891	1.903	1.902	1.617	1.943	1.876	1.853	1.873	1.844	1.869
Vac(A)	0.067	0.117	0.141	0.337	0.316	0.082	0.091	0.046	0.105	0.166
X _{Na}	0.184	0.194	0.879	0.850	0.127	0.259	0.253	0.290	0.272	0.252

Appendix 10.	Chemical	compositions	of white	micas.
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xyge

n = number of analyses

Assemblages:

Ky-Bt-Pl rock (coarse grained)

94. Ms-Ky-Bt-Chl-Pl 95. Ms-Ky-Bt-Chl-Pl

Ky-Crd-Sil rock (zone)

99. Ms-Crd-Bt-Ky-Sil-Pl-Qtz 100. Ms-Crd-Bt-Ky-Sil-Pl-Qtz 101. Ms-Crd-Bt-Ky-Sil-Pl-Qtz 102. Ms-Crd-Bt-Ky-Sil-Pl-Qtz 103. Ms-Crd-Bt-Ky-Sil-Pl-Qtz

Oam-Crd-Ky rock (zone) 96. Pg-Crd-Pl

97. Pg-Crd-Pl

Grt-Qtz rock 98. Ms-Grt-Pl-Qtz