GEOCHEMICAL EVOLUTION OF THE NYIRAGONGO VOLCANO (Virunga, Western African Rift, Zaire)

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Strongly undersaturated foiditic lavas (nepheline—melilitites, nephelinites, leucitites) from the active Nyiragongo volcano have been analysed for trace elements (Sc, V, Cr, Co, Ba, REE, Hf, Ta, Th and U) by instrumental neutron activation analysis.

The overall low Sc and Cr contents and low Mg/Fe ratios suggest that the Nyiragongo parent magma evolved from less undersaturated mantle magmas (olivine—nephelinitic, basanitic) by high-pressure pyroxene fractionation. Pyroxene fractionation seems to be the main cause of the pronounced petrological differences between the adjacent Nyiragongo and Nyamuragira volcanoes.

The distribution of incompatible trace elements within the melilite—nephelinitic magma column was strongly affected by migrating CO_2 -rich fluids. This is most clearly reflected in the unusually large fractionation of U from Th (Th/U = 4.5 down to 0.15), and the high and variable REE contents of the carbonated nepheline—melilitites (bergalites).

Major element differentiation within the magma column was dominated by melilite fractionation in the early stages. Later on, subtraction of pyroxene \pm magnetite) and flotation of leucite (\pm nepheline) gave rise to the formation of highly foiditic porphyritic leucite—nephelinites and nepheline—leucitites.

Key words: volcanic rocks, geochemistry, trace elements, alkaline rocks, nephelinites, African Rift

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Introduction

The Nyiragongo volcano is located in the Virunga (or Birunga) area of the Western African Rift system. The Virunga field consists of eight major volcanoes (Fig. 1) and numerous satellitic eruption centres. None of these volcanoes appears to be older than 2.5 Ma (Pouclet 1980 a), though there is mounting evidence for older Neogene volcanic series in the Virunga area (Kampunzu *et al.* 1983). These older series are probably related to the mildly tholeiitic and alkaline volcanism of the South Kivu field. The still active Nyiragongo and Nyamuragira are the

youngest volcanoes (Upper Pleistocene, Recent). The Nyiragongo has long been famed for its semi-permanent lava lake.

The mineralogy and petrochemistry of the potassic and foiditic Virunga volcanics have been studied extensively since the pioneering work of Holmes and Harwood (1937). Results of earlier studies have been summarized by Bell and Powell (1969), Sahama (1978) and Pouclet (1980 a, b). Major element compositions and succinct petrographical descriptions of a large number of samples have been compiled by Denaeyer *et al.* (1965) and Denaeyer (1972).

The Nyiragongo lavas are quite distinct from those of the other Virunga volcanoes, including the adjacent Nyamuragira. Almost all Nyiragongo lavas are strongly undersaturated feldspar-free nepheline-melilitites and nephelinites; of the feldspathoids, leucite is generally far less abundant than nepheline (Sahama 1968, 1978). The primitive lavas from the other volcanoes are plagioclase-bearing, melilite-free basanites (s.l.), usually with leucite as the sole feldspathoid mineral. In many respects, the Nyiragongo rocks are intermediate between the main Virunga rock series and the ugandites, mafurites and katungites from the Toro-Ankole (Fig. 1) region (Bell and Powell 1969; Pouclet 1980 a, b).

Although the petrogenesis of alkaline undersaturated magmas is not yet fully understood, high-pressure experimental studies have shown they can be generated by small degrees of partial melting of mantle peridotite enriched in H_2O and CO_2 (e.g., Bultitude and Green 1967; Brey and Green 1975, 1977; Wyllie 1979; Edgar and Condliffe 1979). Crustal assimilation, often invoked in older petrogenetic models, is not necessary for the generation of potassic undersaturated magmas (see summary of older hypotheses in Bell and Powell, 1969). In any case, the uniform and relatively low 87Sr/86Sr isotopic ratios of the Nyiragongo lavas (0.7045 \pm 10: Bell and Powell, 1969; 0.70457-0.70479: Vollmer and Norry, 1983) preclude any substantial crus-



Fig. 1. (Left) Location of the three main volcanic fields within the Western African Rift System. Ruled areas are lakes. (Right) Distribution of the eight major volcanoes within the Virunga field.

tal contamination.

To further constrain petrogenetic and magmatic evolution models for the Nyiragongo, we have analysed 46 well-documented samples for trace elements (Sc, V, Cr, Co, Ba, REE, Hf, Ta, Th and U). This work supplements the less comprehensive trace element studies by Bell and Powell (1969) and Mitchell and Bell (1976). The two main objectives of the present study were: (i) to test the detailed magmatic evolution model proposed by Sahama (1973, 1978); (ii) to establish why the two active the Nyiragongo and the Nyamuragira volcanoes are characterized by different rock series.

Samples and analytical method

Most of the samples were collected during several Belgian and Finnish expeditions to the Nyiragongo in the fifties. Sample selection for this study was guided by the late Professor Th. G. Sahama. The sample locations are listed in Table 1. Samples from the inner main cone are well represented because the well defined lava stratigraphy makes it possible to trace the magTable 1. Sample location.¹

- 1. Inner main cone.²
- Inner Pit strata: LL59(MN. Lava lake 1959), VM706(MN), VM701(MN), VM511(MN), VM742(MN)
- Lava beds from inner walls of main cone:
 VM210(BE), VM269(BE), VM235(BE), VM569(BE),
 VM268(BE), VS216(MLN), VM574(LN).
- Dykes from inner walls of main cone: VM273(BE), VM234(BE), VM220(MLN), VM274(LN), VM213(LN), VM212(LN), VM241(LN).
- Lavas from flanks of main cone.² VM12(MLN), FEAE93(MLN), FEAE89(LN), S97(NL).
- Shaheru satellite crater. VM394(MN).
- Baruta satellite crater. S94(ON), S96(MN), S88(MN), VM362(MN), S90(LN).
- Lava flows east of Goma-Rutshuru road. S55(ON), S59(ON), S69(OL), S68(MN), FEAE83(MN), S6(MN), S40(LN), S43(NL), S45(NL), FEAE85(NL), VM75(NL).
- Lava flows west of Goma-Rutshuru road. S9b(OL), S75(MN), Dn54/101(MLN), S107(LN), S10(LN), S12(LN).
- Reference is made to Denaeyer *et al.* (1965) and Denaeyer (1972) for further details of sample location, succinct petrographic description and major element composition. Rock nomenclature: BE = bergalite; MN = melilite-nephelinite; MLN = melilite-leucite-nephelinite; LN = leucite-nephelinite; NL = nepheline-leucitite; ON = olivine-(melilite)-nephelinite; OL = olivine-leucitite.
- ² Refers to pre-1972 configuration of the inner main cone (Sahama 1978).

S901 S451 S75² LL59² VM121 FEAE89² S941 S691 VM234² VM269² 35.13 37.93 38.51 37.71 41.70 46.50 33.08 34.70 SiO₂ 40.32 40.68 3.71 2.24 3.44 1.75 3.00 3.71 2.64 2.42 2.83 2.71 TiO₂ 17.99 15.45 19.00 12.98 13.04 15.87 13.82 13.09 11.97 11.31 Al_2O_3 6.27 3.44 3.70 7.70 9.09 3.25 5.33 5.17 Fe₂O₃ 5.04 6.53 8.21 5.93 8.55 4.18 6.68 9.79 3.02 FeO 6.92 6.64 5.52 0.30 0.29 0.27 0.25 0.26 0.05 0.21 0.21 0.30 0.29 MnO 4.70 2.73 5.07 4.14 MgO 8.72 8.43 7.29 5.52 6.50 4.52 10.98 10.67 5.84 CaO 14.17 15.23 11.87 13.57 13.86 15.50 17.18 4.57 6.44 Na₂O 3.45 2.15 3.16 3.97 4.31 5.34 4.39 5.64 4.79 8.75 4.29 6.28 2.75 3.60 4.86 4.75 5.68 K₂O 3.22 1.25 0.78 1.90 1.67 1.91 1.44 1.84 1.34 1.49 0.38 P205 0.49 0.35 0.43 0.65 $H_2O +$ 0.68 0.93 1.34 3.94 3.41 2.38 0.16 0.17 0.12 H_2O- 0.24 0.200.48 0.15 1.98 3.16 0.40 0.11 CO_2 99.49 99.72 99.94 99.73 99.60 100.20 99.60 99.26 99.73 Total 100.00 BE MN MN MLN LN NL LN ON OL BE Type³

Table 2. Major element composition of selected Nyiragongo samples.

Ref.: ¹ Denaeyer *et al.* (1965), ² New analyses, Analyst: D. Coetermans, Leuven University, ³ Rock type, see footnote in Table 1.

Table	3.	Trace	element	composition	(ppm)	of	Nviragongo	lavas
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		Sc	v	Cr	Co	Ba	La	Ce	Nd	Sm	Eu	Tb	Yb	Lu	Hf	Та	Th	U
LL59	MN	3.9	209	7	44.5	2388	177	357	133	17.3	5.05	1.62	2.90	0.38	4.4	16.2	22.2	8.8
VM706	MN	4.6	273	46	44.2	2037	184	367	135	18.2	5.30	1.69	3.10	0.41	4.8	16.5	21.4	8.2
VM701	MN	3.6	247	52	39.4	1650	153	310	119	15.8	4.62	1.42	2.50	0.35	3.9	16.2	17.6	6.9
VM511	MN	4.6	235	7	35.6	1730	153	306	114	15.4	4.44	1.43	2.50	0.33	4.1	15.0	18.1	7.7
VM742	MN	4.4	278	8	42.2	1955	165	322	122	16.8	4.90	1.47	2.63	0.38	4.5	15.5	18.7	7.3
VM210	BE	8.7		10	40.8	3211	264	504	183	24.8	7.27	2.38	4.30	0.66	6.7	27.6	36.3	22.
VM269	BE	7.8	350	11	37.0	2089	214	413	151	21.2	5.86	1.84	3.34	0.49	5.3	21.7	28.6	12.4
VM235	BE	7.9		10	38.5	6400	265	498	173	22.5	7.13	2.40	4.11	0.59	5.9	26.9	35.0	238.
VM569	BE	6.6	484	9	41.4	2164	239	470	175	23.9	6.76	2.06	3.54	0.49	5.2	23.9	28.7	12.6
VM268	BE	15.5	335	157	40.2	1697	177	336	141	18.4	5.07	1.58	2.86	0.39	5.6	15.6	20.6	10.7
VS216	MLN	5.2		8	46.7	1604	145	286	114	15.6	4.44	1.43	2.34	0.43	4.8	12.1	14.7	4.0
VM574	LN	1.8	169	60	33.2	1793	139	273	97	13.4	3.89	1.14	2.26	0.33	3.6	15.2	17.1	3.8
VM273	BE	6.4	320	16	39.3	2385	235	449	163	21.6	6.17	2.01	3.68	0.47	5.3	20.4	30.0	18.2
VM234	BE	9.6	401	28	43.3	1740	194	384	144	19.9	5.71	1.75	3.08	0.40	5.0	19.5	23.6	12.3
VM220	MLN	2.3	161	≤ 2	27.1	1782	136	268	101	13.2	3.89	1.21	2.41	0.33	3.2	14.2	16.1	5.6
VM274	LN	5.1	204	11	36.6	2207	181	336	112	14.6	4.30	1.30	2.86	0.42	4.8	16.4	27.6	6.5
VM213	LN	5.1	206	9	29.5	2139	172	325	115	14.2	4.13	1.23	2.69	0.36	4.5	16.1	24.5	16.5
VM212	LN	1.50		8	17.2	3755	209	361	112	13.7	4.01	1.33	2.90	0.49	4.6	23.6	37.9	12.3
VM241	LN	1.25		7	14.3	3300	179	312	91	11.4	3.41	1.14	2.58	0.37	3.8	20.9	34.0	20.
VM12	MLN	7.1		8	50.9	2380	151	308	127	17.8	5.19	1.67	2.68	0.44	5.9	12.8	15.2	8.0
FEAE93	MLN	2.9	207	4	35.8	2133	161	324	119	16.4	4.73	1.46	2.80	0.41	4.3	16.4	18.8	6.7
FEAE89	LN	3.2	285	7	36.5	1584	166	322	115	15.6	4.33	1.32	2.60	0.36	3.5	16.6	22.0	8.7
S97	NL	1.32		8	14.8.	3163	184	325	95	11.8	3.52	1.13	2.54	0.43	4.2	20.9	33.9	7.9
VM394	MN	12.9		≤ 3	38.1	3223	265	514	191	26.8	7.80	2.59	4.45	0.79	7.4	25.9	32.6	14.4
S94	ON	20.9		365	50.0	1316	128	255	100	14.2	4.19	1.38	2.53	0.40	6.2	11.0	15.0	5.5
596	MN	10.9		71	44.5	1989	192	379	146	19.6	5.72	1.80	2.82	0.46	5.0	18.1	19.9	9.4
S88	MN	3.6		3	41.8	1787	157	309	118	15.9	4.59	1.53	2.48	0.37	3.3	15.8	18.1	7.3
VM362	MN	9.3		37	41.9	2260	201	371	138	19.0	5.69	1.87	3.41	0.54	5.7	18.9	17.3	11.5
S90	LN	1.43		≤ 2	17.3	2304	148	262	85	10.5	3.12	1.01	2.28	0.42	3.4	15.0	23.9	9.6
\$55	ON	12.2		68	51.5	1243	122	246	101	14.8	4.30	1.42	2.42	0.44	6.0	10.2	13.6	3.9
\$59	ON	29.8		268	49.0	1050	109	222	94	13.9	4.07	1.31	2.08	0.36	7.2	9.4	12.1	4.3
S69	OL	33.4		166	53.3	1070	80	168	74	11.5	3.33	1.13	1.94	0.31	7.1	8.8	9.0	2.2
S68	MN	9.2	386	12	36.0	1930	208	413	159	22.2	6.12	1.98	3.57	0.49	6.2	21.5	25.5	19.
FEAE83	MN	4.4		16	41.9	2032	206	426	158	21.3	6.10	1.89	3.28	0.49	4.0	23.6	27.4	11.0
S6	MN	5.0		9	45.6	2414	173	343	131	17.6	5.11	1.67	2.91	0.45	4.7	16.2	18.9	7.1
S40	LN	4.5		≤ 4	46.3	2140	171	345	134	17.8	5.27	1.70	2.97	0.46	5.3	15.8	19.5	5.9
S43	NI.	3.5		11	42.6	2002	146	294	115	16.0	4.62	1.52	2.56	0.45	4.6	13.2	16.0	5.4
\$45	NL	93		28	42.0	2470	142	274	104	14.3	4.21	1.39	2.48	0.39	6.2	14.1	19.3	5.5
FFAF85	NL	5.8	209	18	34 1	1502	120	234	88	12.1	3.43	1.04	2.14	0.31	5.0	12.2	16.7	5.0
VM75	NL	4.8	202	15	36.7	1797	141	269	99	13.6	3.97	1.20	2.40	0.35	4.5	15.5	19.8	5.7
S9h	OL	18.1		265	47.6	1473	114	227	92	13.6	3.88	1.33	2.28	0.37	6.5	12.9	14.0	5.2
\$75	MN	7.4	385	25	42.5	1932	196	390	152	20.6	5.98	1.86	3.16	0.42	4.9	16.5	22.0	9.6
DN54101	MLN	4.3	295	11	44.2	1972	169	338	135	17.7	5.09	1.57	2.85	0.41	4.8	14.3	18.0	4.7
S107	LN	4 2	275	4	42.8	2090	164	319	123	16.8	4.87	1.56	2.70	0.46	4.9	14.7	18.7	5.6
S10	LN	4 4		53	45.0	2154	169	339	131	17.7	5.20	1.68	2.97	0.49	5.3	15.9	20.0	6.8
S12	LN	4.4		≤ 3	44.2	2188	169	339	130	17.7	5.18	1.74	3.02	0.48	5.4	16.0	19.6	6.5

matic evolution in time. The lavas from the 1972 and 1977 eruptions have not been studied, but one may assume that they are not too different from the lava-lake material sampled in 1959 (sample LL59).

The nomenclature of the Virunga rocks has long been complicated by the use of local and inappropriate names. The rock terminology adopted in the present paper is based on the simplified and revised classification proposed by Sahama (1978) and Pouclet (1980 a, b). Rock names and abbreviations are listed in Table 1. As recommended by Sahama (1978), the name bergalite has been retained for the calcite-rich and olivine-poor Nyiragongo nepheline-melilitites. The olivine-nephelinites and olivineleucitites are not common lava types and are mainly associated with old, secondary eruption centres.

The trace elements were determined by instrumental neutron activation analysis. The induced gamma-ray activities were measured with large Ge(Li) detectors and small high-resolution low-energy photon detectors (Hertogen and Gijbels 1971). Samples (800 mg for whole rocks, 150 mg for mineral separates) were irradiated for seven hours at a neutron flux of ca. 2.10^{12} n.cm⁻².sec⁻¹ in the Thetis reactor of the Institute of Nuclear Sciences, Rijksuniversiteit Gent. Vanadium was determined via ⁵²V (T_{1/2} = 3.8 min) after five minutes irradiation. The samples were analysed relative to a second-



Fig. 2. Alkali and silica content of Nyiragongo samples studied in this work. See text and Table 1 for rock nomenclature and abbreviations. Almost all lavas from the adjacent Nyamuragira volcano plot within the dotted area (Pouclet 1980 a). The Nyiragongo and Nyamuragira are characterized by distinctly different rock series. However, the olivine-nephelinites and olivine-leucitites (circled) fall off the main Nyiragongo trend. In most respects (see also Figs. 3, 5–7) they are intermediate between the Nyamuragira lavas and the most undersaturated Nyiragongo lavas (bergalites, melilite-nephelinites).

ary in-house alkali basalt standard that had been repeatedly calibrated against international reference rocks.

Main chemical fractionation trends

The major element geochemistry of the Nyiragongo rocks has been discussed at length elsewhere (Sahama 1973, 1980; Pouclet 1980 a, b; Pouclet *et al.* 1983) and will not be exhaustively treated in this paper. For ease of reference, the compositions of 10 representative samples are presented in Table 2. Table 3 shows the trace element data for all the samples analysed in the present study.

The compositional differences between the Nyiragongo and Nyamuragira lavas are well illustrated by the Na₂O + K₂O versus SiO₂ diagram (Fig. 2). The Nyiragongo rocks are highly alkaline with K₂O/Na₂O = 1. The melilite-bearing rocks in particular are strongly undersaturated and the most undersaturated bergalites have rather high H₂O and CO₂ contents. The CO₂ derives mainly from calcite in the groundmass. Only leucite-rich rocks have more than 41 % SiO₂.

Attention is drawn to the position of the



Fig. 3. Variation in Sc content and Mg/(Mg + Fe) molar ratio in the Nyiragongo lavas. Total iron was used because the Fe^{3+}/Fe^{2+} ratios are extremely variable and affected by secondary oxidation. Most of the Nyiragongo samples are moderately to strongly fractionated. The low Sc contents point to important pyroxene ± garnet fractionation. Crosses are three primitive anakaratrites from the Nyamuragira (Smet 1980).

olivine-(melilite)-nephelinites and olivine-leucitites in Fig. 2. These rocks deviate from the main Nyiragongo trend and are closer in composition to the least evolved Nyamuragira anakaratrites, basanites and limburgites. In almost all respects the olivine-nephelinites and olivine-leucitites fill the compositional gaps between the Nyiragongo and Nyamuragira lavas (Figs. 2, 6, 7). It is obvious that they provide some sort of a petrogenetic link between the two adjacent volcanoes.

It must be emphasized that, apart from the olivine-nephelinites and leucitites-, the Nyiragongo rocks are moderately to strongly fractionated, as evidenced by the low Mg/(Mg + Fe)ratios (Fig. 3) and low Sc (Fig. 3) and Cr (Table 3) concentrations. The low Sc contents impose important constraints on petrogenetic models for the Nyiragongo. They indicate extensive fractionation involving clinocpyroxene or garnet, or both. However, the rather limited fractionation of heavy from light rare earth elements (Figs. 4—6) rules out the importance of garnet fractionation. (See section on crystal fractionation below and papers by Irving (1978) and Irving and Frey (1978) for crystal/liquid partition coefficient data.)

Like almost all undersaturated alkaline rocks, the Nyiragongo lavas are characterized by high concentrations of the incompatible lithophile trace elements Rb, Sr, Zr, Nb, Ba, light REE, Hf, Ta, Th and U (this work; Bell and Powell 1969; Mitchell and Bell 1976). Enrichment of light over heavy REE in the Nyira-





Fig. 4. Chondrite-normalized REE patterns of selected Nyiragongo lavas. Bergalite VM235 has the highest and nepheline-leucitite FEAE85 the lowest REE abundances of all samples analysed. Note that the patterns of the Nyiragongo samples are virtually parallel.

Fig. 5. Chondrite-normalized REE patterns of selected Nyiragongo lavas (this work) and Nyamuragira ankaratrites (Smet 1980). Note the similarity between Nyamuragira lavas and the Nyiragongo olivine-leucitites.

gongo lavas is more pronounced than in lavas from the other Virunga volcanics (Mitchell and Bell 1976; Smet 1980), though the chondritenormalized La/Yb ratios (30—45; Fig. 6) are much lower than those observed in the rocks from the Toro-Ankole region (La/Yb_N = 80— 200; Mitchell and Bell 1976).

Two conspicuous features of the REE geochemistry of the Nyiragongo lavas need to be stressed and should be accounted for by any viable magmatic evolution model. First, the ma-



Fig. 6. Summary of REE systematics of the Nyiragongo samples. La/Yb ratios are chondrite-normalized. Four aspects of the data should be emphasized: (1) La/Yb ratios within the main Nyiragongo series are rather constant in spite of large variations in absolute REE abundances; (2) absolute REE contents tend to decrease with increasing SiO₂ contents (compare with Fig. 2); (3) absolute REE abundances of the bergalites are very variable, though bergalites have very similar major element compositions; (4) the olivine-nephelinites and olivine-leucitites bridge the gap between the Nyamuragira lavas and the Nyiragongo main series.

jority of the lavas have virtually parallel REE patterns in spite of the large differences in absolute REE abundances and in degree of differentiation (Figs. 3—6). Second, the concentration of REE and other incompatible trace elements tends to decrease with increasing K_2O content and increasing degree of differentiation (Figs. 3 and 7). Hence the potassium enrichment cannot simply be ascribed to crystallization of a K-poor mafic mineral phase, as this would lead to concomitant enrichment of incompatible elements.

An amazing result of the present study is the evidence for an unusually large fractionation of U from Th (Fig. 8). We shall argue below that the Th/U fractionation and the REE distribution in the bergalites is strong evidence for fluid fractionation in the Nyiragongo magma column.

The nature of the parental magma

Two alternative petrogenetic models are proposed as explanations of the strongly undersaturated character of the Nyiragongo rocks: 1. The majority of the magmas derive from a primary melilite-nephelinitic or nepheline-melilititic parental magma formed by minor partial melting of mantle peridotite enriched in H₂O and CO_2 (e.g. Brey and Green 1975, 1977). The occurrence of less undersaturated olivine-nephelinites and olivine-leucitites, and the differences between the Nyiragongo and Nyamuragira magmas in general could then be interpreted in terms of different degrees of melting or variable amounts of H₂O and CO₂ in the respective source regions (e.g., Frey et al. 1978). 2. The Nyiragongo and Nyamuragira volcanoes evolved from very similar 'basanitic' (s.l.) parental magmas. However, the parental magmas underwent a different fractionation history, which in the case of Nyiragongo led to increasing silica-undersaturation and enrichment of alkalies, incompatible trace elements and volatiles. This model has the advantage over the first one that it accounts naturally for the apparent absence of really primitive magmas from the most undersaturated Nyiragongo lavas (Fig. 3).

We shall now argue that the second model is the more plausible one in the light of experimental studies on high-pressure liquidus phase relations in undersaturated magmas in the presence of H₂O and CO₂ (olivine-nephelinites: Bultitude and Green 1967; olivine-melilitite: Brey and Green 1975, 1977). These studies showed that aluminous pyroxenes are the major near-liquidus phases at pressures above ca. 20 kbar. Upon fractionation of pyroxenes (+ garnet) a Nyamuragira-type primitive magma would differentiate towards strongly undersaturated alkaline compositions characterized by relatively low Mg/(Mg + Fe) ratios and low Sc and Cr contents, as is indeed observed for the Nyiragongo bergalites and melilite-nephelinites (Fig. 3; Table 3). In any case, the low Sc contents of Nyiragongo lavas point to the important role of pyroxene \pm garnet fractionation, because only these two phases have high (> 2)crystal/liquid partition coefficients for Sc (Irving 1978; Irving and Frey 1978).

The pyroxene fractionation model has been tested with major element least-squares mass balance calculations, using a computer program based on the mathematical model developed by Stormer and Nicholls (1978). The pyroxene and garnet compositions were taken from Bultitude and Green (1967) and Brey and Green (1977). The differentiation from olivine-nephelinites and olivine-leucitites to bergalites and melilitenephelinites could be modelled quite satisfactorily by fractionation of 25-35 % clinopyroxene plus 3-6 % orthopyroxene. The smallest residuals were generally obtained when using the pyroxene compositions of Bultitude and Green (1967). This appears reasonable because the average composition of the magma was closer to that of nephelinites than melilitites during this fractionation stage. Given the sensitivity of the calculations to changes in the pyroxene compositions, it must be admitted that the relative proportions of clinopyroxene and orthopyroxene are not rigidly constrained. Inclusion of garnet in the calculations did not consistently improve the results. Nevertheless, the increase in the La/Yb ratio during differentiation from olivine-nephelinites to bergalites and melilite-nephelinites (Fig. 6) appears somewhat too large to be explained by 30 to 40 % pyroxene fractionation alone. Fractionation of relatively small amounts of garnet, characterized by high heavy REE partition coefficients (Irving and Frey 1978) might be required to remedy this problem.

It is obvious that 30-40 % pyroxene fractionation only partly accounts for the high REE contents observed in some bergalites and melilitenephelinites (Figs. 4-6). In the next section we shall present evidence that these high abundances of REE and other incompatible trace elements are due to fluid fractionation in the Nyiragongo magma column.

With the data available it is not possible to establish whether the olivine-nephelinites represent the Nyiragongo primary magma or whether they in turn evolved from a basanitic magma by pyroxene subtraction. The extent of prior pyroxene fractionation seems to be limited by the high Cr (> 350 ppm) and Sc (> 30 ppm) contents of some olivine-nephelinites. This could imply that the parental magma of the Nyiragongo and Nyamuragira volcanoes is not exactly the same.

Sahama (1973, 1978) and Pouclet (1980 a) have suggested that the Nyiragongo and Nyamuragira lavas might ultimately derive from very similar parental magmas. Our analysis of the data support this view, though we invoke different differentiation processes. Sahama attributed the difference in chemistry between the two volcanoes to the concentration of an alkali-rich, carbonated fluid phase in the Nyiragongo magma. This model partially accounts for the low SiO₂ contents — merely the result of passive SiO₂ depletion — but leaves the Mg/Fe, Sc and Cr fractionation unexplained. We shall argue below

that fluid fractionation did indeed occur in the Nyiragongo magma column. However, the fluid phase was probably not strongly alkaline, as the $H_2O + CO_2$ rich bergalites generally have lower Na₂O and K₂O contents than the volatile-poor melilite-nephelinites (Fig. 2).

Crystal and fluid fractionation in the Nyiragongo magma column

The lava stratigraphy inside the Nyiragongo main cone offers valuable clues to the magmatic evolution during the main eruptive phases of this volcano (Tazieff 1966; Sahama 1978). The oldest exposed strata are made up of bergalites cut by dykes and covered with lavas of (melilite)leucite-nephelinitic and nepheline-leucititic composition. Because the younger lavas in this sequence are the most fractionated ones (Figs. 2 and 3), the magmatic evolution of the Nyiragongo can possibly be largely interpreted in terms of fractionation of a bergalitic or melilitenephelinitic parent magma. This parent magma itself derived from a primary olivine-nephelinitic or basanitic magma, as argued in the preceding section.

Note that the most recent lavas from the inner pit and lava lake (Table 1) are less differentiated than the older leucite-nephelinites and nepheline-leucitites. This might imply that the magma chamber or the top of the magma column beneath the Nyiragongo has been replenished at least once with less fractionated magma.





Fig. 7. La and K contents of the Nyiragongo lavas. REE abundances tend to decrease with increasing K₂O contents owing to flotation of leucite and nepheline.

Fig. 8. Fractionation of U from Th in the Nyiragongo lavas is unusually large, and is hard to explain by simple crystal fractionation. The uranium enrichment is attributed to fluid fractionation in the magma column. The horizontal line drawn at Th/U = 3.8 represents the commonly accepted value for primitive mantle.

Before evaluating the crystal fractionation models we should asses the role of fluids. Consideration of fluid fractionation is prompted by the high H₂O and CO₂ contents of the bergalites (Fig. 2), and especially by the unusually large Th/U fractionation (Fig. 8). Since none of the major phenocryst phases of the Nyiragongo lavas selectively concentrates either Th or U (Table 4), the Th/U fractionation is hard to explain by crystal fractionation. A far more plausible explanation is very strong partitioning of uranium into a fluid phase as a result of formation of stable chloride, fluoride, and carbonate complexes (Langmuir 1978). In any case, the Th/U fractionation must be attributed to abnormal behaviour of U, because the ratios of Th to other incompatible trace elements such as La and Ta are within the ranges typical of alkaline basic rocks (La/Th = 6-10; Th/Ta = 1-1.5; Wood et al. 1979).

Fluid fractionation also makes it easier for us to understand why the concentrations of incompatible trace elements (REE, Hf, Ta, Th, U) correlate rather poorly with the major element composition and degree of fractionation of the samples (Figs. 2, 6-8). This is strikingly illustrated by the almost twofold variation in REE abundances in the bergalites, which show only minor differences in major element composition (Figs. 2 and 9). The high REE contents of carbonatites (e.g., Loubet et al. 1972; Eby 1975) and the experimental data (Wendlandt and Harrison 1979) prove that REE are strongly fractionated into carbonate-rich fluids. Hence we conclude that the distribution of REE in the Nyiragongo magma column has been controlled to a large extent by a carbonated phase whose presence is still reflected in the high calcite content of the bergalites.

Our data support Sahama's view (Sahama 1973, 1978) that the bergalites only represent the gas-laden, carbonated top of the Nyiragongo magma column. Their occurrence is therefore confined to the initial, explosive eruptive phases. The parent magma was only moderately enriched in volatiles, and it was closer in composition to the most undersaturated melilite-nephelinitic lavas (SiO₂ = 35-36 %; H₂O + CO₂ = 2-4 %).

Crystal fractionation in the Nyiragongo magma chamber or magma column has also been modelled with major element least-squares mass balance calculations, using the phenocryst compositions from Sahama (1978).

The calculations resolved two main fractionation stages. The first one applies to the differentiation of the bergalites and melilite-nephelinites. Compositional changes in the lavas are mainly



Fig. 9. Variation of atomic Ca/Si, Al/Si and Si/Mg ratios of the Nyiragongo lavas. Relative abundances of Ca, Al, Mg and Si are controlled by crystal fractionation involving melilite, clinopyroxene, nepheline and leucite. Melilite subtraction largely accounts for the decreasing Ca/Si and increasing Al/Si and Si/Mg ratios of bergalites and melilitenephelinites. The superimposed fractionation vectors show the effect of clinopyroxene and nepheline subtraction (—) and nepheline and leucite flotation (+) upon differentiation of melilite-nephelinites to leucite-nephelinites.

due to crystallization of melilite (5-10 %), olivine (2-4 %) and apatite (2-3 %). In addition, up to 5 % of nepheline had to be either removed from or added (by flotation) to the melts to satisfy mass balance constraints. During this fractionation stage, the SiO₂ content — recalculated to volatile free compositions - of the derivative magmas increased slightly (Fig. 2). The Ca/Si ratios decreased (Fig. 9) as a result of melilite fractionation, whereas the Si/Mg ratio increased (Fig. 9) and the Mg/(Mg + Fe) ratio decreased (Fig. 3) as a result of olivine fractionation. The increase in the Al/Si ratio (Fig. 9) is largely accounted for by melilite and olivine crystallization, implying that nepheline fractionation was not generally important. The abundant nepheline phenocrysts apparently crystallized in situ and remained suspended in the melt owing to their relatively low density (ca. 2.66 g/cm^{3}).

The second fractionation stage applies to the melilite-leucite-nephelinites, leucite-nephelinites and nepheline-leucitites. Fractionating phases are clinopyroxene, Ti-rich magnetite, apatite, nepheline and leucite. The effect of clinopyroxene, nepheline and leucite fractionation during this stage is illustrated by fractionation vectors superimposed on Fig. 9. Flotation of low density leucite (ca. 2.47 g/cm³) was rather important during this stage. Flotation of nepheline was required to model the compositions of the samples containing abundant rounded nepheline phenocryst aggregates (Sahama 1978). However, the scarcity of samples having Al/Si ratios greater than 0.5 (Fig. 9) implies that the high modal abundance of nepheline in the 'nephelineaggregate lavas', leucite-nephelinites and nepheline-leucitites is only partly due to crystal flotation.

Fractionation of clinopyroxene (up to 30 %) and magnetite (up to 8 %) turned out to be more important than the phenocryst abundances would suggest. In any case, the significant decrease in the Mg/(Mg + Fe) ratio and Sc abundances (Fig. 3) requires removal of clinopyrox-

Table 4. Phenocryst/whole rock concentration ratios.

	CPX ¹	MEL ²	NEPH ³	LEUC ⁴
Sc	9.2	0.03	0.02	0.007
Cr	1.8	0.02	_	_
Co	1.1	0.86	0.02	0.009
Ba	0.05	0.10	0.21	1.1
La	0.35	0.33	0.011	0.015
Ce	0.51	0.35	0.011	0.016
Nd	0.83	0.41	0.009	0.017
Sm	1.0	0.36	0.010	0.018
Eu	1.0	0.40	0.025	0.014
Tb	1.1	0.28	0.023	
Yb	0.94	0.095	0.023	0.018
Lu	1.0	0.082	0.028	_
Hf	2.8	0.10	0.046	0.07
Та	0.25	0.001	0.025	0.01
Th	0.11	0.023	0.022	0.02
U	0.05	0.024	0.033	0.03

¹ Clinopyroxene from sample FEAE93, ² Melilite from sample FEAE83, ³ Nepheline from samples FEAE86 and FEAE88, normalized to whole rock data for FEAE89, ⁴ Leucite from sample FEAE85.

ene. The increase in the SiO_2 content (Fig. 2) during this fractionation stage is largely the result of leucite flotation and magnetite removal. The combined effect of clinopyroxene removal and leucite flotation is clearly reflected in the increasing Si/Mg ratios, the decreasing Ca/Si ratios and the levelling off of the Al/Si ratios at a value of about 0.5 (Fig. 9).

Whereas the trace element abundances in the bergalites and melilite-nephelinites were mainly controlled by fluid fractionation, crystal fractionation was a more important process during the second differentiation stage. Melilite, clinopyroxene, nepheline and leucite mineral separates were analysed for trace elements to permit us to estimate crystal/liquid partition coefficients, (Table 4). Althought not analysed, apatite is known to have high partition coefficients (> 5) for all the REEs (Nagasawa 1970; Watson and Green 1981).

Nepheline and leucite partition coefficients are very low for all the elements measured except Ba. Flotation of nepheline and leucite, together with removal of apatite, largely accounts for the generally lower concentrations of incompatible trace elements in the leucite-nephelinites and nepheline-leucitites (Figs. 4—6). Flotation of leucite is further reflected in the high K and Ba contents of leucite-rich lavas.

The REE partition coefficients for clinopyroxene are between 0.35 and 1.1, which are rather high values. With partition coefficients so close to unity, absolute REE abundances of derivative lavas should not vary much with the degree of clinopyroxene fractionation. However, extensive clinopyroxene fractionation appears to be responsible for the higher La/Yb ratios (> 40; Fig. 6) of samples VM212, VM241 and S90, since these three samples also happen to have the lowest Sc contents, the lowest Mg/(Mg + Fe) ratios (Fig. 3) and highest Si/Mg ratios (they plot off scale in Fig. 9).

Summary and concluding remarks

From the low Mg/Fe ratios and low Sc and Cr contents of the strongly undersaturated Nyiragongo nepheline-melilitites (bergalites) and melilite-nephelinites we infer that they derived from less undersaturated olivine-nephelinitic or basanitic (*s.l.*) mantle magmas by high-pressure (≥ 20 kbar) pyroxene fractionation. Our data hence support earlier suggestions by Sahama (1978) and Pouclet (1980 a, b) that the adjacent Nyiragongo and Nyamuragira volcanoes ultimately evolved from very similar primary mantle magmas.

The main Nyiragongo lava types were formed by progressive low-pressure fractionation of a nepheline-melilititic (or melilite-nephelinitic) parent magma. Differentiation was dominated by melilite crystallization at the earlier stages. Later on, removal of clinopyroxene and flotation of leucite (\pm nepheline) led to the formation of highly fractionated porphyritic leucitenephelinites and nepheline-leucitites.

The bergalites represent the volatile-enriched top of the melilititic magma column. Migrating

carbonate-rich fluids affected strongly the distribution of incompatible trace elements. Fluid fractionation is most clearly reflected in the unusually large Th/U fractionation and the high and variable REE abundances of the bergalites.

The essential aspects of Sahama's (1973, 1978) petrogenetic model for the Nyiragongo are corroborated by trace element data; he did not, however, apparently fully appreciate the importance of high- and low-pressure pyroxene fractionation.

Our data provide some constraints on the interpretation of the unusual Pb isotopic variations in the Nyiragongo lavas (Vollmer and Norry 1983), but we admit that the model presented here might need considerable revision once the causes of the Pb isotopic variations are fully understood.

Vollmer and Norry consider that the Pb isotopic variations are a feature inherited by the magmas from a heterogeneous mantle source. The source heterogeneity itself is presumed to be due to a ca. 500-Ma-old mantle metasomatic event. This interpretation can only be correct if (i) the seven samples analysed for Pb isotopes represent as many separate batches of mantle magmas; or if (ii) the radiogenic Pb was highly concentrated in an accessory mantle mineral that survived melting and failed to equilibrate with the melts (the interpretation preferred by Vollmer and Norry). The first assumption is within the realm of possibility, though it is rather surprising that every magma batch apparently experienced the same, somewhat complex fractionation history (Figs. 2-3). The second assumption does not seem to be very plausible, especially because the most radiogenic nepheline-leucitites and leucite-nephelinites are the end-products of a complex differentiation sequence and therefore had the least chance of retaining accessory mantle minerals.

Explanations in terms of contamination by old U-rich crystal phases should perhaps be reinstated. The evidence for extensive low-pres-

sure fractionation undoubtedly implies that the Nyiragongo magmas have been trapped at crustal levels for extended periods. Whatever the case, solution of the Nyiragongo lead puzzle will have to await additional Pb isotopic analyses of well-documented samples. It would be of great interest to learn if Pb isotopic compositions correlate with rock type, and if there is a link between Pb isotopic variations and the

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